

Development of a Simple Method for Determination of Anti-cancer Component of Indole-3-carbinol in Cabbage and Broccoli

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Abstract Indole-3-carbinol is a potential anti-cancer agent, and it has been found rich in *Brassica* vegetables. However, there was few good extraction method of indole-3-carbinol in cabbage and broccoli. In this work, the wavelength and the ration of mobile phase in determination method, as well as different temperature and pH values in extraction method were all investigated some cultivars of cabbage and broccoli were collected and validated the determination method. A sensitive and rapid high-performance liquid chromatography method has been established and used for the determination of indole-3-carbinol in cabbage (*Brassica oleracea* L. var. *capitata*) and broccoli (*Brassica oleracea* L. var. *italica*). The method was proved to be sensitive, selective, rapid and reproducible, with a good recovery of 99.25%. There was a fast retention time (4.88 min) and good linearity ($R^2=0.9991$) in the system. Meanwhile, significant differences of indole-3-carbinol contents were detected among all the materials ($P<0.05$). Finally a simple extracting method of indole-3-carbinol was established, at the same time, the determination method was shown accurate, reliable and stable. So the determination method could be proposed application in medical and industrial area.

Keywords: indole-3-carbinol, cabbage, broccoli, HPLC, anti-cancer

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1. Introduction

Indole-3-carbinol (I3C) is the second product of hydrolysis glucobrassicin (3-indolylmethyl glucosinolate), which is found in cruciferous vegetables such as broccoli, cabbages, brussels sprouts, collard greens, kale, cauliflower, kohlrabi, mustard greens, radishes, rutabaga and turnips [1,2]. Epidemiological and clinical studies suggest that indole-3-carbinol is a potential agent in blocking tumour initiation, oestrogen metabolism, inhibition of cell proliferation and induction of apoptosis [3,4,5,6,7]. So far, many studies have shown indole-3-carbinol is a bioactive component that plays an important anti-cancer role against breast [8], liver [9], prostate [10], lung and colon cancers [11,12]. Thus, indole-3-carbinol has been widely studied and has proved to be a potentially useful agent in cancer prevention.

Some researchers have reported that indole-3-carbinol is abundant in cruciferous vegetables because it is derived from indole-3-glucosinolate (I3G), which is found in cabbage, broccoli, kale, cauliflower, and turnip. It has been proved that consumption of cabbage juice is

beneficial as an antioxidant [13] and as an anti-cancer agent [14,15].

Recently, the determination methods of indole-3-carbinol have been reported by GC-MS [16] and HPLC [17,18,19], which have been found in mouse plasma, mustard and Chinese cabbage [19,20]. Indole-3-carbinol is unstable in cell culture medium and physiological fluids since much of it is converted in the intestine to its metabolites diindolymethane (DIM) and indole carbazole (ICZ) [21]. Thus, the method of detection of indole-3-carbinol requires serious consideration. GC/MS methods require high temperatures, usually resulting in degradation of this component and, consequently, a decrease in sensitivity. HPLC methods are typically used to determine glucosinolate and their secondary products with an appropriate extraction procedure. Thus HPLC is a suitable technique for qualitative and quantitative study of indole-3-carbinol and similar compounds [17,22].

Few reports have described a simple method for the extraction and fast detection of indole-3-carbinol in *Brassica* vegetables. Thus, it is necessary to establish a process for the fast determination of indole-3-carbinol in vegetables. This could be useful for the selection of germplasm and biochemistry.

2. Materials and Methods

2.1. Materials and Reagents

Fifteen cabbage cultivars and ten broccoli cultivars were bred and harvested in autumn 2015. All cultivars, including 15 cabbage varieties and 10 broccoli varieties, were obtained from the Chinese Academy of Agricultural Science (CAAS), Institute of Vegetables and Flowers (IVF). Materials were planted in a field at IVF during August 2015. At maturity, all cabbage and broccoli plants were harvested as soon as possible, and the harvest samples were quickly frozen by liquid nitrogen. All the fresh samples were dried by a freeze drying machine. Finally the dry samples were ground into powder and stored in sealed bags at -20°C .

Indole-3-carbinol was purchased from Sigma-Aldrich (St. Louis, MO, USA) and had a purity greater than 98%. HPLC grade acetonitrile and methanol, as well as a double-distilled water Milli-Q quality water system (Millipore, Bedford, USA), were obtained from Supervision and Testing Center for Vegetable Quality, Ministry of Agriculture (Beijing, China). Sodium dihydrogen phosphates, citric acid, and ethyl acetate were purchased from Beijing chemical works (Beijing, China).

2.2. Extraction of Indole-3-carbinol

Powder samples (0.5 g) were weighed and homogenized with 15.0 mL of sodium dihydrogen phosphate and citric acid buffer solution and shaken for 1.5 hours in a flask, and the solvent ethyl acetate was added to the extraction by shaking for half an hour. The extracted solvent was centrifuged for 10 min at $5500 \times g$ in 50-mL tubes. This procedure was repeated twice. The resulting solvent (EtOAc soluble layer) was evaporated under reduced pressure using a rotavapor (RII, BÜCHI, Switzerland) at 30°C . The residue was then re-dissolved in 10 mL of methanol. The extract was filtered through Agela (China) No. 0.22- μm (D 13 mm) nylon filter paper and stored at -20°C for HPLC determination.

We prepared a series of sodium dihydrogen phosphate and citric acid buffers with varying pH values of 5, 5.5, 6.0, 6.5, 7.0 and 8.0. These were used to determine an ideal condition for the indole-3-carbinol generation catalysed by myrosinase. Varying concentrations of indole-3-carbinol solution were also designed for investigation of indole-3-carbinol stabilization at pH values of 1.0, 2.0, 3.0, 4.0, 5.0, 6.0, 7.0 and 8.0. Samples collected at 0 h, 1 h, 2 h, 4 h, 6 h, 8 h, 12 h and 24 h from the indole-3-carbinol solutions were analysed.

Temperatures of 25°C (room temperature), 30°C , 50°C and 70°C were used in the experiments, and samples collected at 0 h, 1 h, 2 h, 4 h, 6 h, 8 h and 12 h were analysed. We also used this protocol to determine a condition good for the indole-3-carbinol hydrolysis reaction.

2.3. Method Validation

HPLC was run on an SHIMADZU LC-20A series high-performance liquid chromatography system, equipped with a CTO-20A column oven, SPD-20A UV/VIS detector,

SIL-20A HT auto-sampler, LC-20AD pump, and DGU-20A3 degasser. The separation was carried out on a Waters PAH C_{18} column (250×4.6 mm, $5.0 \mu\text{m}$). The mobile phase was optimized using water and acetonitrile, and varying ratios of water to acetonitrile were used: 70 to 30, 60 to 40, 50 to 50, 30 to 70 and 40 to 60. The peak of indole-3-carbinol was investigated and analysed.

The precision was calculated the peak areas of an indole-3-carbinol standard solution at a concentration of 50 mg/L with eight times determination of indole-3-carbinol ($n=8$), the RSD was calculated for showing the precision.

The accuracy of the determination of indole-3-carbinol was also carried out by analysing the peak areas of an indole-3-carbinol standard solution at a concentration of 49.5 ($n=8$), the RSD was calculated for showing the accuracy.

Stability was calculated using the peak area of an indole-3-carbinol standard solution at a concentration of 49.5 mg/L every two hours ($n=8$), which also shown by the RSD.

Recovery was carried out by known cabbage samples spiked with indole-3-carbinol at concentrations of 10.0 $\text{mg}\cdot\text{L}^{-1}$, 12.24 $\text{mg}\cdot\text{L}^{-1}$, 14.52 $\text{mg}\cdot\text{L}^{-1}$, 15.23 $\text{mg}\cdot\text{L}^{-1}$, 15.82 $\text{mg}\cdot\text{L}^{-1}$, 20.43 $\text{mg}\cdot\text{L}^{-1}$, each treatment was extracted and analysed in triplicate ($n=3$), and the results are shown as RSD (%) in Table 1.

Table 1. Recovery of indole-3-carbinol results

Cabbage sample $\text{mg}\cdot\text{L}^{-1}$	Spiking $\text{mg}\cdot\text{L}^{-1}$	I3C		
		Detection mg/L	Recovery (%)	RSD (%)
12.53	10.00	22.01	97.69	1.00
12.52	12.24	24.6	99.31	
12.53	14.52	26.53	98.08	
12.52	15.23	28.11	101.26	
12.52	15.82	27.52	98.48	
12.53	20.43	33.19	100.7	

Linearity was evaluated by injecting indole-3-carbinol standard solutions with concentrations of 3.33 $\text{mg}\cdot\text{L}^{-1}$, 10.0 $\text{mg}\cdot\text{L}^{-1}$, 20.0 $\text{mg}\cdot\text{L}^{-1}$, 50.0 $\text{mg}\cdot\text{L}^{-1}$, 100.0 $\text{mg}\cdot\text{L}^{-1}$ and 130.0 $\text{mg}\cdot\text{L}^{-1}$ in triplicate ($n=3$).

The determination of detection limits (LODs) and quantification limits (LOQs) was performed by analysis of the decrease in the concentration of the standard solutions. Detection limits were estimated based on a 3-to-1 signal-to-noise ratio by injecting standard solutions injection in quadruplet. The quantification limits were calculated according to the formula $\text{LOQ (RSD} < 5\%) = 10/3 \text{ LOD}$ [22,23].

2.4. Application in Cabbage and Broccoli

A total of 15 cabbage varieties and 10 broccoli varieties were collected for extract of indole-3-carbinol based on above method, and the residue was diluted in methanol.

Selectivity was determined by comparing the chromatograms of cabbage samples with those of the standard solutions. Peak identification was achieved by analysing the retention times as they appeared after the injection of indole-3-carbinol standard solutions and

cabbage samples separately under identical conditions. In the end, the concentration of indole-3-carbinol among cabbage and broccoli varieties was recorded.

2.5. Statistical Analysis

Indole-3-carbinol contents, standard error (SD), and validation data were all analysed by SPSS 13.0. After one-way ANOVA, the Duncan test was performed to test the significance of the differences between mean values using a probability level of $p < 0.05$ [16,24]. The test data are shown as the mean values \pm SD.

3. Results and Discussion

3.1. Wavelength and Eluant Selectivity of Indole-3-carbinol

Wavelengths of 200-400 nm were scanned, and indole-3-carbinol detection was carried out at an

absorption wavelength of 279 nm using a standard solution in methanol (Figure 1), which was consistent with previous literature [17,19]. The chromatogram showed a good peak with no interfering signal (Figure 2).

HPLC reverse-phase C₁₈ has been widely used for gluconisolate determination [13,25,26,27,28]. A Waters PAH C₁₈ (250×4.6 mm, 5.0 μ m) column was suitable for the separation of indole-3-carbinol. Mobile phases of methanol and water or acetonitrile and water were used at ratios 30 to 70, 40 to 60, 60 to 40 and 70 to 30. The mobile phase of water and acetonitrile proved to be sensitive and able to rapidly identify and separate indole-3-carbinol followed the procedure described above. Finally, the mobile phase was determined, which included water (pump A) and acetonitrile (pump B) according to the following programme: pump B was initially set at 40% acetonitrile and then linearly increased to 60% over 7 min, after which it returned to 100%. It was then allowed to equilibrate for 13 min at a flow rate of 0.80 mL·min⁻¹. The column temperature was 32°C, and the injection volume was 10 μ L·min⁻¹.

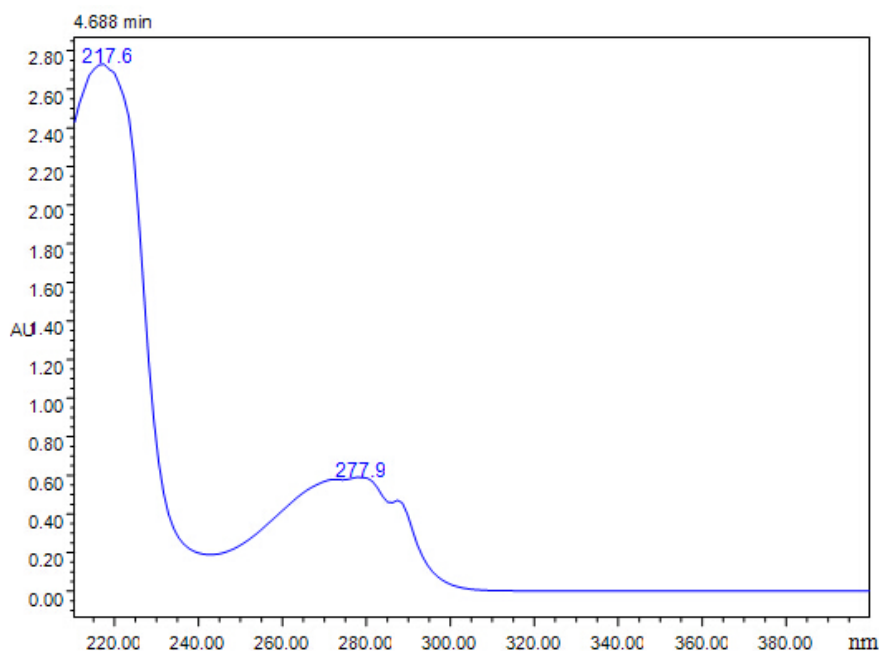


Figure 1. The UV wavelength scanning result of indole-3-carbinol from 200 to 400 nm

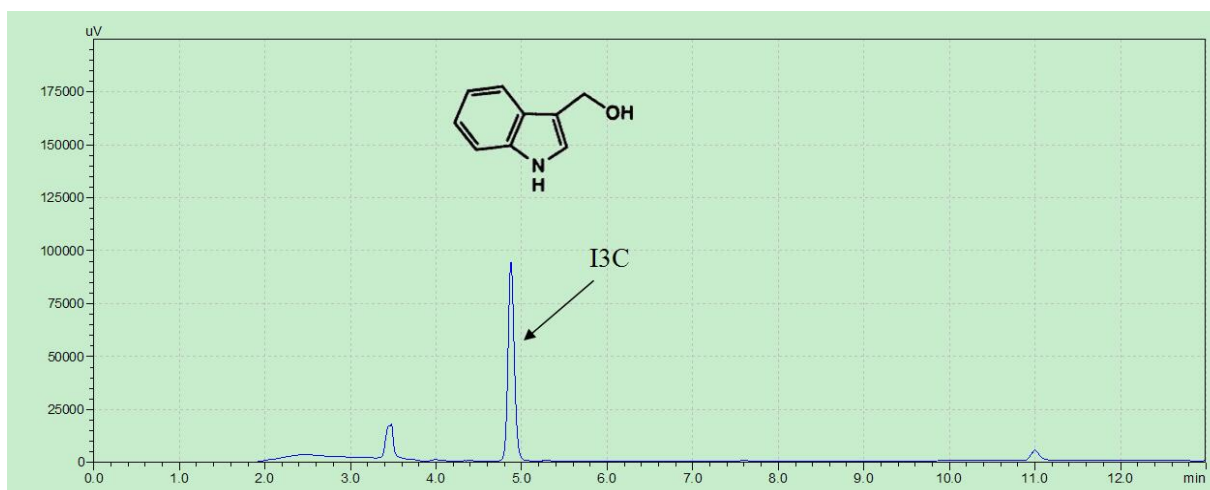


Figure 2. HPLC chromatograms of indole-3-carbinol in standard

3.2. pH Conditions for Hydrolysis of Indole-3-carbinol

In this process, pH and temperature are the main factors that affect the generation of indole-3-carbinol catalysed by myrosinase. Thus, we examined different pH conditions to optimize the extraction method for indole-3-carbinol.

From the data in Figure 3, we found that there was a higher generation of indole-3-carbinol at pH 8.0. The results indicated that pH values between 5 and 6.0 were not good for indole-3-carbinol generation, and more indole-3-carbinol can be produced at a pH between 6.5 and 8.0. The highest amount of indole-3-carbinol was produced at pH 8.0. This result indicated that this alkaline environment was good for myrosinase-catalysed indole-3-carbinol generation and that this condition was good for the stabilization of indole-3-carbinol (Figure 3). Previous studies showed that pH 7.4 was also an acceptable condition for the extraction of indole-3-carbinol; however, this condition is not the most ideal. According to this optimized study (Figure 3), we found that the alkaline condition described herein was good for the generation of indole-3-carbinol catalysed by myrosinase [17,19].

3.3. Temperature for Hydrolysis of Indole-3-carbinol

Another factor affecting indole-3-carbinol generation is temperature; because indole-3-carbinol is not stable, fluctuations can cause aggregation into 3,3'-diindolylmethane (DIM) under strongly acidic conditions. indole-3-carbinol and its condensation product 3,3'-diindolylmethane (DIM)

have both been shown to be promising anti-tumour agents from *Brassica* vegetables [21,29,30].

Thus, we designed a trial to test the stability of indole-3-carbinol under different temperature conditions over twelve hours. According to our investigation, room temperature (25°C) was an ideal condition to stabilize indole-3-carbinol, as indole-3-carbinol would degrade or convert into other components within one hour at temperatures over 50°C (Figure 4). As shown in Figure 4, we also found that indole-3-carbinol was relatively stable at 30°C in the first six hours but degraded or changed after six hours. Thus, temperatures over 25°C were not suitable for the storage and stability of the indole-3-carbinol solution.

In the process of indole-3-carbinol hydrolysis, myrosinase played a key role in the composition and amount of indole-3-carbinol in *Brassica*, similarly with sulforaphane hydrolyzed from glucoraphanin [31,32]. Myrosinase is affected by genotype, pH, temperature, ferrous ion, zinc and magnesium concentration, which is still a complex problem in the stability of myrosinase [19,33]. In our study, we focused on the stability of indole-3-carbinol in the dominant factors of pH and temperature. There are still more works to do for investigation of myrosinase.

3.4. HPLC System

Compared with previous reports, the HPLC conditions described in this study allowed a sensitive and fast determination of indole-3-carbinol with a retention time of 4.8 min [11,19] and a reasonable run time of 13.0 min.

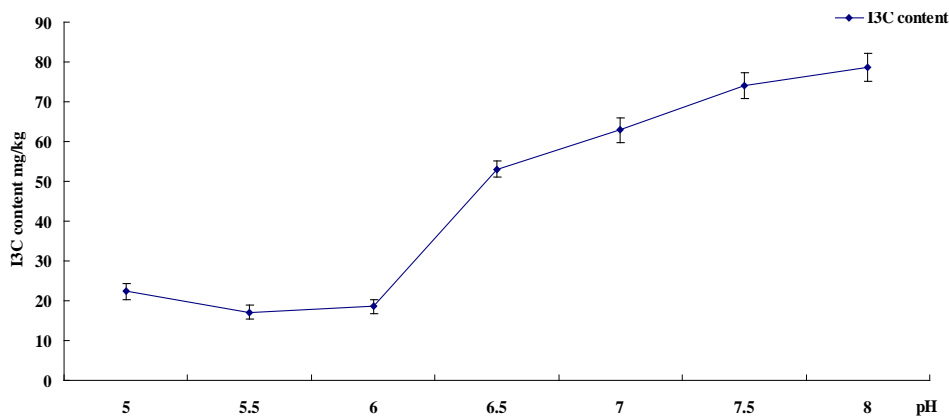


Figure 3. Hydrolysis product of indole-3-carbinol at different pH conditions

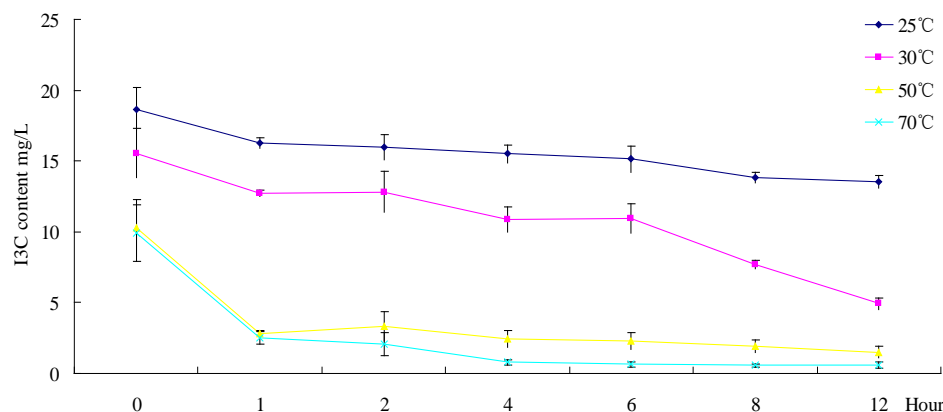


Figure 4. Stability of indole-3-carbinol in different temperature

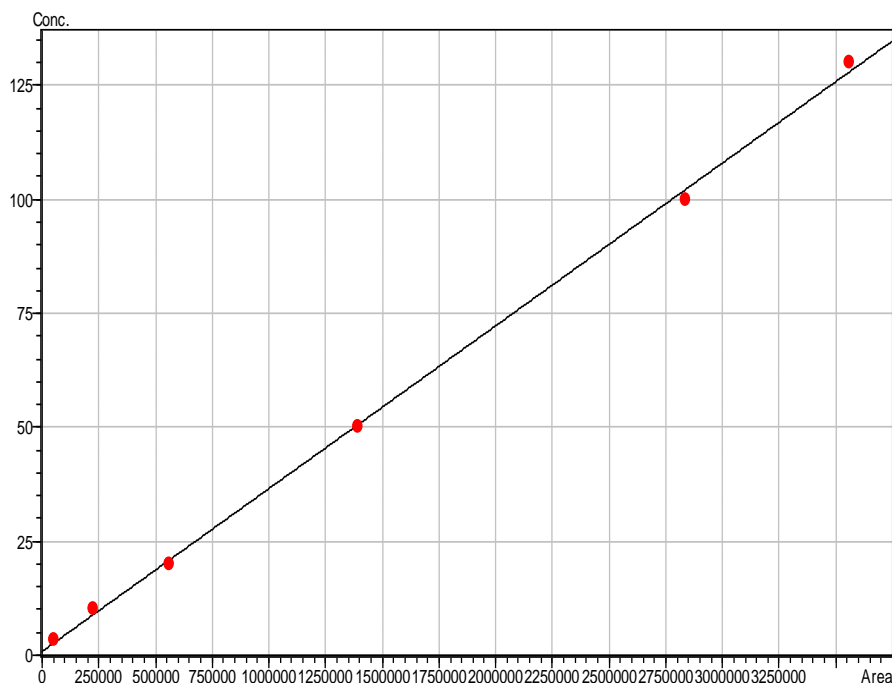


Figure 5. Calibration curve of indole-3-carbinol standard concentrations ranging from 3.33 mg·L⁻¹ to 130 mg·L⁻¹

From Figure 5, we could find there was a good linearity, the equation was $Y = 3.57 \times 10^{-5} X + 0.85$, $R^2 = 0.9991$. According to calculation, we recorded the LOD of indole-3-carbinol was 0.02 $\mu\text{g}\cdot\text{L}^{-1}$, meanwhile the LOQ was 0.07 $\mu\text{g}\cdot\text{L}^{-1}$.

The precision of the instrument was calculated by analysing the peak areas of an indole-3-carbinol standard solution at a concentration of 50 mg/L; the RSD (n=8) of precision was 0.44% (Peak areas: 1410362, 1408512, 1408202, 1405097, 1402286, 1391297, 1400870 and 1400660).

The accuracy of the determination of indole-3-carbinol was also determined by analysing the peak areas of an indole-3-carbinol standard solution at a concentration of 49.5; the RSD (n=8) of accuracy was 0.42% (Peak areas: 1384515, 1381645, 1377171, 1374232, 1373870, 1371607, 1369721 and 1367734).

Stability was calculated using the peak area of an indole-3-carbinol standard solution at a concentration of 49.5 mg/L every two hours (n=8); the average RSD (%) was 0.32 (Peak areas: 1384515, 1381645, 1377171, 1373870, 1371607, 1373924 and 1374538).

Cabbage samples (12.52 mg·L⁻¹, I3C) were spiked with different concentrations of indole-3-carbinol following Table 1. From these results, we see that the average recovery of this method was 99.25% (RSD=1.00%), which suggested a good recovery based on this method.

3.5. Application on Cabbage and Broccoli

Fifteen cabbage cultivars and ten broccoli cultivars were extracted and validated by the HPLC method. The determined quantities of indole-3-carbinol are summarized in Table 2. All the materials were analysed and found to have a bioactive component of indole-3-carbinol, and there was a strong indole-3-carbinol peak observed (Figure 6).

In each cultivar, we detected a component of indole-3-carbinol; the average indole-3-carbinol level of 15 cabbage cultivars was 28.85 mg·kg⁻¹ DW. The highest indole-3-carbinol content was in C238 (87.88 mg·kg⁻¹

DW), while the lowest was in C575 (6.62 mg·kg⁻¹ DW). The indole-3-carbinol content in 15 cabbage cultivars ranged from 6.62 to 87.88 mg·kg⁻¹ DW. The values from every cultivar of cabbage were significantly different ($P < 0.05$). So we could infer that the genotype affect the hydrolysis product of indole-3-carbinol, which also provided indirect and more evidence for diversity of glucosinolate and hydrolysis products derived from glucosinolate, such as sulforaphane [24,34]. In this study, the cabbage C238 was demonstrated to a higher resource in indole-3-carbinol, which might provide a natural material for plant breeding, nutrition and anti-cancer research.

Table 2. The contents of indole-3-carbinol in cabbage and broccoli cultivars

Cabbage cultivars	Content of I3C (mg·kg ⁻¹)	Broccoli cultivars	Content of I3C (mg·kg ⁻¹)
C238	87.88 ± 0.12 a	B2	65.47 ± 0.32 a
C459	68.36 ± 0.39 b	B33	64.71 ± 0.29 a
C208	45.51 ± 0.44 c	B6	50.38 ± 0.36 b
C222	44.07 ± 2.60 c	B7	48.19 ± 1.25 c
C202	31.18 ± 0.65 d	B1	35.99 ± 1.87 d
C543	30.25 ± 1.09 d	B8	22.25 ± 2.03 e
C497	23.43 ± 1.37 e	B10	20.36 ± 0.52 e
C200	20.14 ± 0.57 f	B12	15.03 ± 0.12 f
C8	14.94 ± 0.56 g	B22	12.17 ± 0.45 g
C576	14.55 ± 0.60 g	B28	8.88 ± 0.73 h
C475	13.76 ± 0.37 g		
C544	12.98 ± 0.85 gh		
C185	10.92 ± 0.55 h		
C580	8.21 ± 0.88 i		
C575	6.62 ± 2.59 i		

Note: the content of indole-3-carbinol is presented as the mean ± SD (n=3); different letters indicated significant differences at the level of 5% based on Duncan's comparison in row.

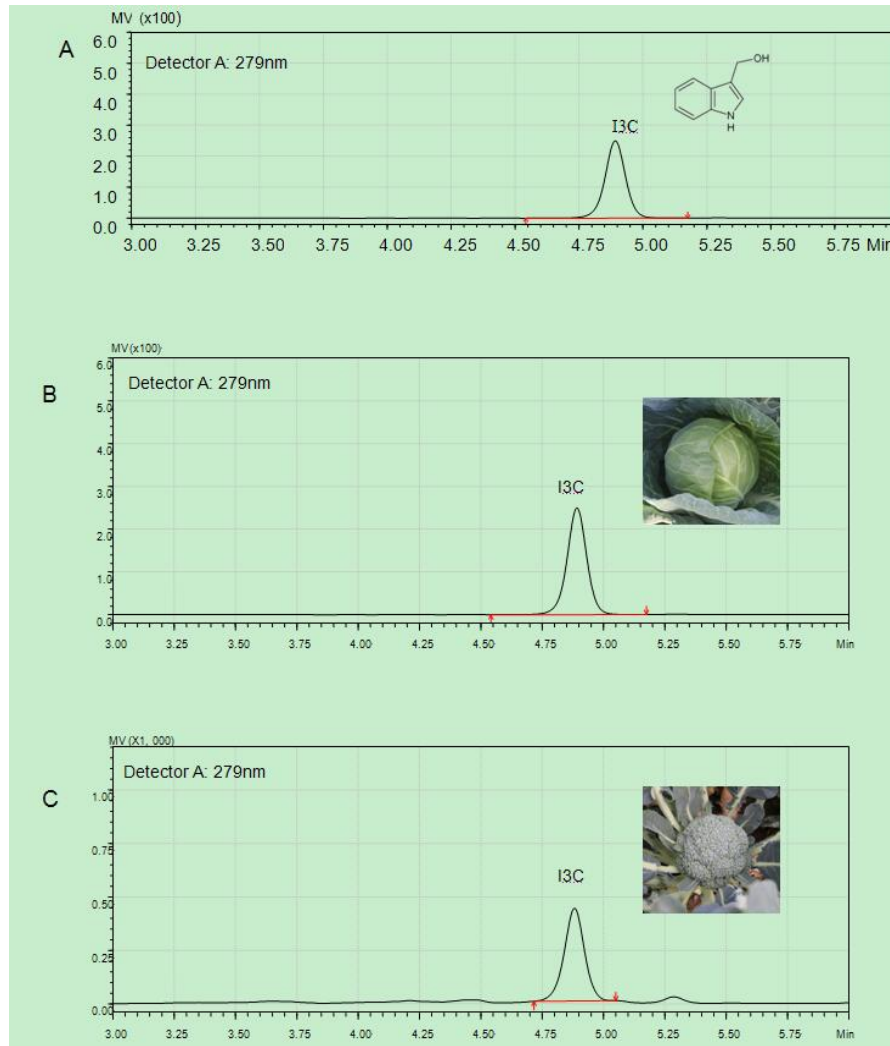


Figure 6. Chromatogram of indole-3-carbinol in standard and samples of cabbage and broccoli

Compared with the cabbage determination of indole-3-carbinol, there was a smaller variation ranging from 8.88 to 65.47 $\text{mg}\cdot\text{kg}^{-1}$ DW among broccoli lines. The average indole-3-carbinol content in broccoli was 34.34 $\text{mg}\cdot\text{kg}^{-1}$ DW, and every cultivar of broccoli was also significantly different ($P < 0.05$). The variation of indole-3-carbinol detected in broccoli was smaller than that in cabbage, which might suggest that the genetic background of indole glucosinolate in cabbage was wider than in broccoli.

The results in this study proved that the method could successfully extract indole-3-carbinol from cabbage and broccoli. The HPLC system was proved a good condition for determination of indole-3-carbinol, validating previous studies [17, 19]. All the materials including cabbage and broccoli were both detected with the anti-cancer component indole-3-carbinol. The average concentration of indole-3-carbinol in cabbage was higher than that in broccoli. However there was a smaller range of variation in broccoli, which might suggest cabbage is a good material for studying the mechanism of indole-3-carbinol in genetic analysis and metabolism. In other studies, Chinese cabbage (0.63 $\text{mg}\cdot\text{kg}^{-1}$ DW) was determined to have a higher content of indole-3-carbinol than cabbage and broccoli, and mustard had a lower content (0.13 $\text{mg}\cdot\text{kg}^{-1}$ DW) [19,35].

In this study, we proved that cruciferous vegetables are rich in indole-3-carbinol, and genotype of *Brassica*

affected the amount of indole-3-carbinol in hydrolysis process. And a diet rich in cruciferous vegetables, such as cabbage and broccoli, is helpful for human health and protection against cancer [36].

4. Conclusion

In this study, we provided a simple and rapid method for the extraction and determination of indole-3-carbinol in Brassica plants with a good recovery, such as cabbage, broccoli, Chinese kale, kohlrabi, and kale. The genotype of Brassica affected the amount of indole-3-carbinol in hydrolysis process. And pH 8.0 was helpful for extracting of indole-3-carbinol, meanwhile the temperature should not be more than 30°C. In fact, the method is good for the fast determination of indole-3-carbinol in medical and industry fields.

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