

Effect of Rice-based Fat Substitute on Gelation of Myofibrillar Proteins

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Abstract This study was designed to investigate the effect of rice-based fat substitute (FS) on heat-induced gelation of myofibrillar proteins (MP) from chicken breast muscle. The secondary structure, calorimetric and rheological attributes (α -helix, β -sheet, heat flow and G') of FS and MP mixtures were measured during gel formation. The results indicated that the addition of FS led to easier denaturation of myosin but delayed denaturation of actin. The α -helix content in the MP-FS mixture was lower than that of MP, whilst β -sheet content in the MP-FS mixture was higher than that of MP when heating temperature was higher than 60 °C, indicating that the addition of FS could promote MP molecules unfolding and aggregating at higher temperature. The G' value of the MP-FS mixture was higher than that of MP during heating. The initial gelling temperatures of the MP sample and the MP-FS sample were 42 °C according to G' curves. The hardness value of the MP-FS gel was higher than that of the MP gel at temperature over 60 °C, and reached the maximum value at 75 °C. Scanning electron microscopy showed that FS changed the microstructure of MP gel. It was concluded that the addition of FS promoted MP molecule unfolding, aggregating and gelling at heating temperature over 60 °C, but FS did not change the initial gelling temperature of MP molecules and the optimal gel-forming temperature.

Keywords: myofibrillar proteins, fat substitute, gelation, gel properties

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1. Introduction

The heat-induced protein gelation is generally known as a process that the native protein denature and their molecules unfold during heating, then the unfolding molecules aggregate and form gel matrix [1]. Myofibrillar proteins (MP) can form heat-induced gel [2]. Heat flow curves of MP measured by DSC may contain 2 or 3 peaks of denaturation temperature because of different experimental conditions [3,4,5,6]. MP may generally denature at 50-54, 57 and 63-66 °C. MP molecules are unfolded and aggregated during heating accompanying with the decrease of α -helix content and the increase of β -sheet content [7,8]. Liu observed that α -helix content of fish myosin (the major component of MP) decreased from 85% to 38% when heated from 30 °C to 80 °C at 1 °C/min [9].

MP gel properties are affected by many factors including such additives as carrageenan, soybean protein, starch and transglutaminase. Amako [10] reported that carrageenan had no effect on the thermal stability of MP from chicken breast but increased water holding capacity of the MP gel. Chin and co-workers [11] studied the effects of soy protein and transglutaminase on rheological properties of pork MP gel. Hong and Chin [12] studied the

effects of alginate on the gel hardness of pork MP. Fat substitute (FS) is a key ingredient of low-fat meat surimi products. Rice-based fat substitute is the hydrolyzate of rice. The gelling process of rice-based FS has been reported by Yang and Xu [13]. The applications of rice-based FS in meat surimi products have been reported by Hsu [14] and Ju [15]. Few data are available on its effect on MP gelation process and gel properties. The objective of the present study was to investigate the effects of rice-based FS on the denaturation temperature, secondary structure, gel rheological and textural properties of MP from chicken breast muscle, to explore the effects of the FS on the gelation of MP.

2. Materials and Methods

2.1. Materials

Six-week-old commercial AA broilers were purchased from a commercial meat company.

2.2. Extraction of MP

MP was extracted from broiler breast muscle according to the procedure of Xiong et al. [16]. MP precipitate was stored at 4 °C and used within 3 days.

2.3. Preparation of Rice-based FS

The rice-based FS was prepared according to the procedure of Yang and Xu [13]. Rice flour was slurred with deionized water to 20% concentration. The rice slurry was gelatinized at 90 °C for 20 min in a water bath with stirring (200 r/min). Then the rice slurry was hydrolyzed by α -amylase until DE 2-3. α -Amylase was inactivated by adding 1 mol/L hydrochloric acid to pH 2.5, then neutralized with 1 mol/L sodium hydroxide 30 min later. The hydrolyzate was spray-dried. Thus, FS was obtained. DE value was determined using the method of Hizukuri, Takeda, Yasuda, and Suzuki [17].

2.4. Measurement of MP Denaturation Temperature

Samples were prepared containing 5 mg MP, 5 mg MP-2.5 mg FS, and 5 mg MP-5 mg FS individually and sealed in the pans. MP thermal stability was measured using a DSC (Pyris1, Perkin-Elmer instrument Ltd), calibrated with indium. Each sample was scanned from 30 to 100 °C at a rate of 10 °C/min. The temperature and enthalpy were recorded during heating.

2.5. Measurement of MP Secondary Structure

The contents of α -helix and β -sheet were measured using a circular dichroism instrument (Chirascan, Applied Photophysics Co., UK). The parameters were: quartz cell 1 cm, Temperature program conditions 30-80 °C by 2 °C/min. Each sample was scanned from 190 to 250 nm at a rate of 100 nm/min. The buffer was used as control. The MP sample concentration was 16 μ g/mL, whilst the MP-FS sample contained 16 μ g/mL MP and 8 μ g/mL FS.

2.6. Measurement of Dynamic Rheological Properties

Rheological properties were measured using a rheometer (MCR302 rheometer, Anton Paar Ltd., Austria). The measurement conditions were: 50 mm diameter parallel plate, gap 0.5 mm, frequency 0.1 Hz, and strain 2%. The samples were heated from 30 to 75 °C at a rate of 1 °C/min. G' were automatically recorded during heating. MP sample concentration was 20 mg/mL, whilst the MP-FS sample contained 20 μ g/mL MP and 10 μ g/mL FS.

2.7. Measurement of Textural Properties of the Heat-induced Gel

MP solutions (30 mg/mL) and MP-FS solutions (30 mg/mL MP, 15 mg/mL FS) were heated from 30 to 45, 50, 55, 60, 65, 70, 75 and 80 °C at 1 °C/min in water bath, respectively, and kept the temperature until a total time of 50 minutes; then cooled to room temperature, and placed

at 4 °C for 16 hours. The gels were used to determine the effects of heating temperature on gel textural properties.

To determine the effect of storage time on gel textural properties, the same concentration MP and MP-FS solutions were heated from 30 to 75 °C and kept the temperature for 5 minutes. The gels were then cooled to room temperature and stored at 4 °C for 20, 40, 60, 80, 100, 120, 140, 160, and 180 minutes, respectively.

Textural properties of the heat-induced gel were determined using a textural analyzer (TA.TX.Plus textural analyzer, Stable Micro Systems Ltd) with TPA program. The measurement conditions were: probe P/6, pre-test speed 5.0 mm/s, test speed 1 mm/s, post-test speed 5.0 mm/s, and distance 5 mm.

2.8. Scanning Electron Microscopy

MP gel (40 mg/mL), MP-FS gel (40 mg/mL MP, 20 mg/mL FS), and MP-FS gel (40 mg/mL MP, 40 mg/mL FS) were prepared and placed at 4 °C for 16 hours. The gel was fixed in 5% glutaraldehyde, washed in phosphate buffer, fixed in 1% osmic acid, dehydrated with ethanol, and dried by freeze-drying. The microstructure of gel was observed under a scanning electronic microscope (XL-30w/Inca, Philips Ltd.) at accelerating voltage of 15 KV.

2.9. Statistical Analysis

All the experiments were triplicated. Data were analyzed by correlation analysis and multiple comparisons using SPSS 17.0.

3. Results and Discussion

3.1. The Effect of FS on MP Denaturation Temperature

DSC thermograms showed that both MP and MP-FS samples had two distinct endothermic transition peaks. Peak 1 and peak 2 were attributed to myosin and actin, respectively [5,18]. The denaturation temperature of myosin in the MP sample was T_o 54.51 °C, T_p 58.99 °C, T_e 62.86 °C; whilst that of myosin in MP-FS(1:1) sample was T_o 51.96 °C, T_p 57.77 °C, T_e 66.63 °C (Table 1), which indicated that addition of FS destabilized myosin and widened the denaturation temperature range of myosin. The T_p values of actin from MP, MP-FS (2:1) and MP-FS (1:1) were 69.80, 74.80 °C and 78.88 °C, which suggested that addition of FS could stabilize actin. This may be due to the interaction between FS and MP. The rice-based FS contains approximately 90% maltodextrin, and the only active group of maltodextrin molecules is hydroxyl. Therefore, the interaction between FS and MP must be hydrogen bonds.

Table 1. Effect of FS on MP denaturation temperature

Samples	Peak 1			Peak 2		
	T_o (°C)	T_p (°C)	T_e (°C)	T_o (°C)	T_p (°C)	T_e (°C)
MP	54.51±0.01	58.99±0.02	62.86±0.02	65.80±0.01	69.80±0.01	72.21±0.01
MP-FS(2:1)	53.58±0.02	57.77±0.00	64.72±0.01	73.05±0.01	74.80±0.01	76.35±0.02
MP-FS(1:1)	51.96±0.0	55.77±0.01	66.63±0.01	77.70±0.01	78.88±0.02	81.25±0.02

Some previous studies have shown that the interaction between meat protein and additives. Yang and co-workers [13] reported that the addition of gelatin led to easier

denaturation of myosin because of the interaction between myosin and gelatin. Chen and co-workers [19] suggested that addition of flaxseed gum (FG) to salt-soluble meat

protein (SSMP) could change the denaturation temperature of the protein because of the electrostatic interaction between SSMP and FG in gel formation. Ma [20] reported that the addition of κ -carrageenan to SSMP led to the disappearance of peak1 and the decrease of T_{peak2} because of the interaction between SSMP and κ -carrageenan. Ramirez and co-workers [21] found that the MP samples exhibited three endothermic peaks in DSC thermogram, and addition of wheat bran led to the decrease in the first T_p and increase in the third T_p . The peak₁ and peak₃ of their DSC thermogram were corresponded to the peak₁ and peak₂ in this present study. Therefore, wheat bran and the FS had the similar effect on MP denaturation.

3.2. The Effect of FS on MP Secondary Structure

The α -helix content decreased from 95.77% to 45.05% when the MP was heated from 30 to 80 °C (Table 2). This

is in accordance with previous studies [7,8,9]. The α -helix content changes of MP-FS and MP samples during heating had the same trend. The initial α -helix content of MP-FS was higher than that of MP (96.89% vs 95.77%). However, α -helix content of MP-FS was 39.70%, lower than that of MP (45.05%) at 80 °C significantly ($p < 0.05$). The β -sheet content increased from 0.20% to 12.65% when the MP was heated from 30 to 80 °C. The β -sheet content of MP-FS transformed faster than those of MP sample from 60 to 70°C ($p < 0.05$). The decline of α -helix content during heating indicated the unfolding of MP molecules, and the increase of β -sheet content meant the aggregating of the unfolding protein molecules [22]. Therefore, FS accelerated the unfolding-aggregating process of MP molecules at temperature higher than 60 °C, whilst it stabled the MP structure at temperature lower than 60 °C. Few reports are available on effect of additives on the MP secondary structure.

Table 2. Effect of FS on secondary structure content of MP during heating

Temperature / °C	α -Helix / %		β -Sheet / %	
	MP	MP-FS	MP	MP-FS
30	95.77±0.81ab	96.89±1.10a	0.20±0.03A	0.10±0.02A
40	94.65±0.53b	95.86±0.05ab	0.30±0.03A	0.21±0.01A
50	89.14±0.05d	91.25±0.06c	1.13±0.04B	0.86±0.01A
60	75.89±0.09e	76.60±0.07e	4.02±0.02C	3.75±0.07C
70	46.73±1.21f	40.85±0.94h	12.05±1.03D	14.13±1.04E
80	45.05±0.56g	39.70±0.96h	12.65±0.12D	14.93±0.06F

Note: Different letters in same row indicate significant difference ($p < 0.05$).

3.3. The Effect of FS on MP Rheological Properties

G' curves of MP and MP-FS samples showed similar trends during heating, and the G' values of MP-FS sample were higher than that of MP sample at the same temperature (Figure 1). G' of MP and MP-FS increased slowly at 42 °C, which was the initial gelling temperature of the MP [16]. Therefore, addition of FS increased the elasticity of the samples, even if the initial gelling temperature of MP did not change.

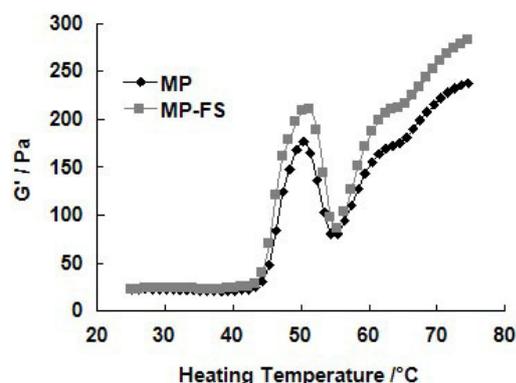


Figure 1. Effect of heating temperature on rheological properties of MP and MP-FS mixtures

Some additives have been reported to affect the rheological properties of muscle protein. The addition of chitosan to SSMP markedly increased the G' values over the entire temperature range [23]. κ -Carrageenan was also found to increase the complex modulus of SSMP [24]. The addition of gelatin to myosin also increased the G' values [25]. These three additives affected G' of MP

because of physical interactions between additives and MP such as electrostatic interaction and hydrogen bond. In the present study, MP and FS interacted by hydrogen bond, which led to higher G' values.

3.4. The Effect of FS on Textural Properties of MP Gel

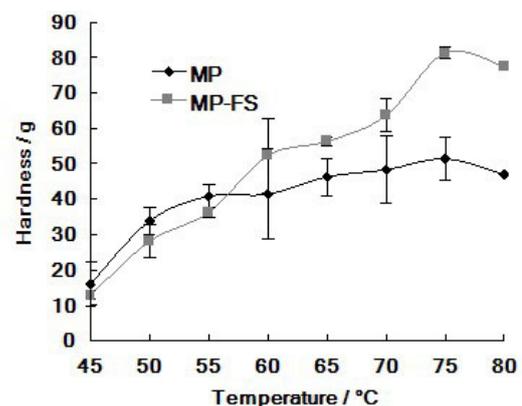


Figure 2. Effect of FS on hardness of MP gel and MP-FS mixed gel

The hardness of both heat-induced MP gel and MP-FS gel increased with temperature. They showed the maximum value of 51.4g and 81.33g at 75 °C, respectively (Figure 2). Because FS (15 mg/mL) could not form gel, it indicated that MP played a critical role in MP-FS gel and there was interaction between MP and FS. The hardness of MP gel was higher than that of MP-FS gel at temperature lower than 55 °C; however the opposite situation appeared when the gels prepared at temperature higher than 55 °C. The hardness of MP-FS mixed gel increased quickly from 55 to 75 °C, whilst the hardness of

MP gel increased slowly. Furthermore, the decrease in the α -helix and the increase in the β -sheet became faster above 60 °C (Table 2). This could be because FS was not soluble at lower temperature and the insoluble FS hindered MP gelling; but the solubilized FS molecules interacted with MP molecules at higher temperature and accelerated MP molecules folding and aggregating, and finally resulted in higher hardness value of the gel.

There have been a few reports about the effects of polysaccharides on the gel hardness of muscle protein. Ma [20] and Verbeken [24] reported that κ -carrageenan could increase the gel hardness of SSMP at proper conditions.

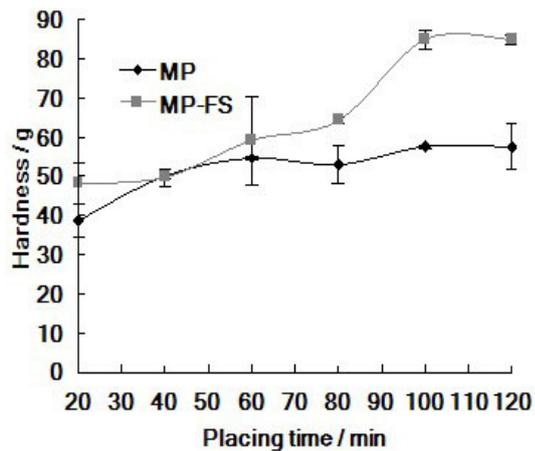


Figure 3. Effect of placing time on hardness of MP gel and MP-FS mixed gel

The effect of FS on textural properties of MP was not only related to the temperature of the gel preparation, but also to the storage time of the gel. The hardness of MP gel stored at 4 °C increased within the first 40 min but not change afterwards. However, the hardness of the MP-FS mixed gel increased significantly till 100 min (Figure 3).

This is because the maltodextrin in FS continued to gel at 4 °C. Yang et al [13] reported that the gel strength of rice-based FS increased at 4 °C within 8 h. Therefore, the interaction between MP and FS and the FS gelation led ultimately to higher hardness for the MP-FS gel than the MP gel. In addition, the presence of MP accelerated the gelation speed of FS.

3.4. The Effect of FS on the Microstructure of MP Gel

The gel matrix formed by MP was fine and uniform (Figure 4a). In MP-FS (2:1) mixed gel, both FS and MP molecules expanded into chain molecules. FS induced MP to forming a thick and strong gel matrix with bigger meshes, then FS molecules attached to the matrix, and entangled into fine gel network in the meshes (Figure 4b). Therefore, SEM demonstrated that the hydrogen bond was formed between FS and MP, which led to the changes of original microstructure of MP gel, and eventually led to changes of the gel properties such as gel hardness. When FS were added into MP samples at a ratio of MP to FS 1:1, the excess of FS existed in particle state and destroyed MP gel network structure (Figure 4c). Therefore, the suitable ratio of MP to FS was 2:1.

Sun [26] reported that adding peanut protein to chicken SSMP could improve the microstructure of the mixed gel and enhance the gel strength. Chin [11] also reported that adding sodium caseinate (SC) or soy protein isolate (SPI) to pork MP could make the gel network compact. Li and Xia [23] have observed that the gel network of SSMP and chitosan mixtures was compact and fine compared with that of SSMP. He considered that there were electrostatic interaction and hydrogen bond between SSMP and chitosan, which contributed to the improvement of the microstructure of the mixed gel.

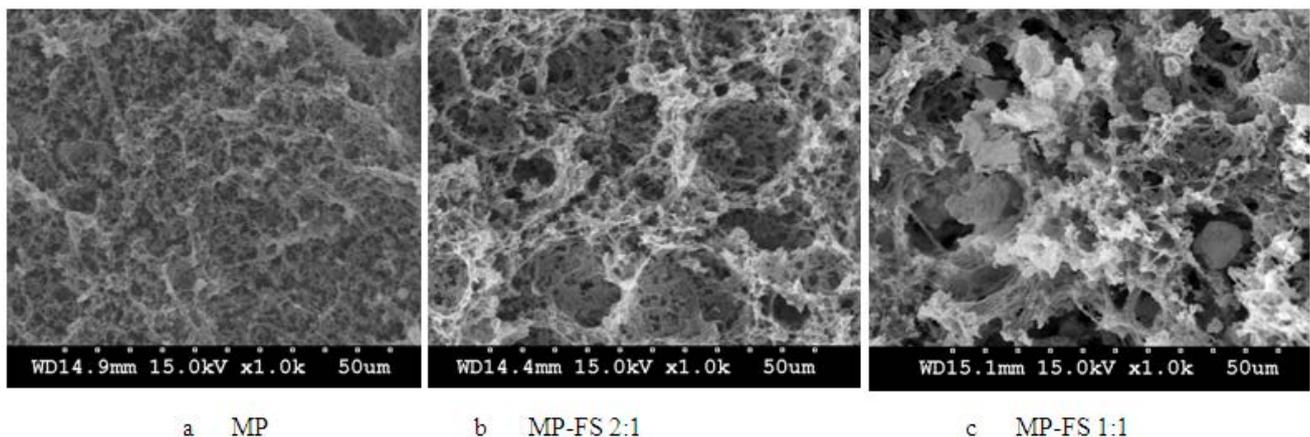


Figure 4. Microstructure of MP gel and MP-FS gels

4. Conclusions

Rice-based fat substitute affected the heat-induced gelation of MP and the microstructure of MP gel because the FS interacted with MP by hydrogen bond. When the temperature was higher than 60 °C, the change rate of MP secondary structure in MP-FS sample was faster than that in MP sample, and the hardness of MP-FS mixed gel was

higher than that of MP gel, which demonstrated that FS improved MP molecules unfolding and the unfolding molecules aggregating and gelling, and strengthened the gel at temperature over 60 °C; whilst FS stabilized the MP structure and weakened the MP gel at temperature lower than 60 °C. Addition of FS didn't change the initial gelling temperature of MP and the optimal temperature of the gel prepared.

Addition of FS changed the original microstructure of MP gel. FS induced MP forming a thick and strong gel

matrix with bigger meshes in the mixed gel, then FS molecules entangled into fine gel network in the meshes. The suitable adding amount of FS to MP was at ratio of MP to FS 2:1. In addition, the presence of MP accelerated the gelation speed of FS.

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