

Effect of Extrusion on the Hydrophilic Antioxidant Capacity of Four Whole Grains

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Abstract Cereal grains are generally subjected to thermal treatment before consumption. However, few data are available on total phenolic content (TPC), degree of starch gelatinization, phenolic acids, and antioxidant capacity of extruded whole grains based on trolox equivalent antioxidant capacity (TEAC), DPPH radical scavenging activity, and ferric reducing antioxidant potential (FRAP). Four whole grains, brown rice, wheat, maize, and barley were examined in this study. The free and bound TPC changed significantly ($p < 0.05$) after extrusion. Moreover, heat and degree of starch gelatinization could affect the bound TPC. In bound fraction, the antioxidant capacity was directly influenced by TPC. Extrusion (110-140°C) can retain or improve the antioxidant capacity in its free and bound fractions. Phenolic acids are stable in their bound form. Results showed that the bound TPC and total phenolic acids have a strong relationship against antioxidant capacity, indicating that TPC is the major antioxidant in the bound fraction.

Keywords: whole grains, total phenolic content, antioxidant capacity, HPLC

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1. Introduction

Cereal grains are traditional Asian staple foods and a major source of daily energy intake. Nowadays, consumers prefer refined grains to whole grains because of the rough texture and unpleasant tastes of whole grains. However, most of the antioxidants are removed during the refining process [1,2]. Epidemiologic studies have shown that the intake of whole grain is associated with positive health benefits, such as lowered risks of chronic diseases like coronary heart diseases (CHD), type II diabetes, and cancer along with reduced mortality risk. This is attributed in part to the phytochemicals located in the edible parts of the whole grain, i.e. the bran, germ, and endosperm [3,4,5,6,7]. Phytochemicals, which include terpenoids, phenols, and alkaloids, are bioactive components of plants that provide many health benefits. It is believed that their antioxidant activities are key to the protection against some chronic diseases [8,9]. Therefore, encouraging people to consume more whole grains on a daily basis appears to be of great importance.

There has been much research into the antioxidant activity of whole grains. Most studies have focused on the use of various organic solvents to fully extract the antioxidants [10,11]. However, some studies have shown that such extraction methods underestimate the antioxidant capacity of whole grains because most of the antioxidants are in an insoluble bound form, i.e. they are bound to cell

wall material [12,13]. Therefore, both free and bound phenolic contents should be considered.

Most of the literature describing the antioxidant capacity of whole grains is focused on the raw materials [11,12,14,15,16,17]. Unlike vegetables and fruits, whole grains can hardly be eaten by humans without being cooked. Therefore, studies on the antioxidant capacity of unprocessed whole grains appear to be inadequate because different processes may have differing effects on the antioxidant content of whole grain. To the best of our knowledge, only a limited number of studies have reported the antioxidant content of processed whole grains [18,19]. Thus, it is necessary to investigate the antioxidant capacity of processed whole grains so that we can accurately evaluate their antioxidant activity.

The objective of this study was to investigate the effects of extrusion (110-140°C) on phytochemical content (i.e. the total phenolic content (TPC), ferulic acid content and *p*-coumaric acid content), degree of starch gelatinization and antioxidant capacity (i.e. DPPH radical scavenging capacity, trolox-equivalent antioxidant capacity [TEAC] and ferric reducing antioxidant potential [FRAP]) of four whole grains (i.e. maize, wheat, brown rice, and barley).

2. Materials and methods

2.1. Materials

Folin-Ciocalteu's reagent, 6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid (Trolox), 2,2'-azino-bis(3-ethylbenzothiazoline-6-sulphonic acid) (ABTS), 2,2-diphenyl-1-picrylhydrazyl (DPPH), Taka-Diastase from *Aspergillus oryzae*, ferulic acid, and *p*-coumaric acid were purchased from Sigma-Aldrich (St. Louis, MO). 2,4,6-Tri(2-pyridyl)-*s*-triazine (TPTZ) was purchased from Aladdin Technology Corporation (Shanghai, China). Methanol and acetonitrile of HPLC grade were purchased from J&K Scientific Ltd (Beijing, China). Other chemicals used in this study were of analytical grade and purchased from Sinopharm Chemical Reagent Co., Ltd. (Shanghai, China).

Whole grain maize, wheat, brown rice and barley were purchased from a local supermarket in Wuxi, China. All whole grains were stored at -20°C before use.

2.2. Samples Preparation

Whole grains were ground (Model FW100 grinder, Tianjin Taisite Instrument Co., Ltd., Tianjin, China) to produce a fine powder that would pass through a 60 mesh sieve. The grounds were stored at -20°C before use.

Extruded samples: Whole grain powder (1 kg) was obtained as described above, and the water content was adjusted to 25%. Extrusion was carried out on a twin-screw extruder (Model POLYLAB, Thermo Fisher Scientific Inc., Massachusetts, USA) at a constant 100 r/min screw speed and 60 g/min mass flow rate. Four barrel temperature profiles were used: 70-80-90-110°C, 70-80-100-120°C, 70-80-100-130°C, and 70-80-110-140°C.

All the samples obtained above were able to be consumed and then ground to produce a fine powder that would pass through a 60 mesh sieve. The grounds were stored at -20°C before extraction.

2.3. Determination of the Moisture Content

For determination of the dry weight (DW), approximately 1 g of raw or processed sample was dried in a drying oven (Model DHG-9203A, Shanghai Yiheng Scientific Instrument Co., Ltd, Shanghai, China) at 105°C for 4 h until the sample reached a constant weight.

2.4. Phenolic Extracts

Extraction of free-soluble and insoluble-bound phenolic compounds was according to the method described by Sosulski [20] and modified in our lab. **Free-soluble phenol:** Whole grain flour samples (10g) were extracted twice with 100 mL of 80% (v/v) aqueous ethanol for 10 min at ambient temperature and centrifuged at 4000 r/min for 10 min at 20°C. The supernatants were collected, combined, and then evaporated under vacuum at 35°C in a rotary vacuum evaporator (Model RV10 basic, Guangzhou IKA Scientific Instrument Co., Ltd., Guangzhou, China) to dryness and reconstituted in 100 mL of water before two extractions with *n*-hexane to remove lipophilic contaminants. **Insoluble-bound phenols:** After the extraction of free-soluble phenols, the residue was digested with 100 mL of 2 M NaOH at ambient temperature for 1 h, and an ultrasonic cleaning instrument (Model KQ2200DE, Kunshan Ultrasonic Instrument Co., Ltd., Kunshan, China) was used during the

hydrolysis. The resultant hydrolysate was acidified to pH 6 with HCl and extracted three times with ethyl acetate. The ethyl acetate fraction was collected and evaporated to dryness under vacuum at 35°C and reconstituted in 100 mL of water, then extracted twice with *n*-hexane to remove lipophilic contaminants.

2.5. Determination of TPC

Briefly, 1 mL of the extract was thoroughly mixed with 1 mL of Folin-Ciocalteu reagent and 3 mL of 7.5% (w/v) sodium carbonate solution [21]. The mixture was allowed to stand at ambient temperature for 2 h. The absorbance was measured at $\lambda = 765$ nm with a Visible Spectrophotometer (Model 722S, Wuxi Keda Intelligent Instrument Co., Ltd., Wuxi, China). TPC was determined by means of a calibration curve prepared by using gallic acid and expressed as microgram gallic acid equivalent (GAE) per gram of whole grain (DW). Data are reported as mean \pm standard deviation (SD) for at least three replications.

2.6. Determination of Starch Gelatinization Degree

The degree of starch gelatinization of the extruded whole grains was determined according to Bhattacharya [22]. Briefly, 3 flasks of 100 mL were prepared which were marked as A₁, A₂ and B, respectively. 1 g of sample was placed in both A₁ and A₂ before 50 mL of water was added into the 3 flasks. In order to thoroughly gelatinize the starch, A₁ was boiled in water for 20 min before it was cooled down to ambient temperature immediately. Then, A₁, A₂, B were respectively added with 5 mL of 5% (w/v) Taka-Diastase which were later water bathed in 38°C for 2 h. Enzymatic hydrolysis was paused by adding 2 mL of 1M HCl. The hydrolysates in A₁ and A₂ were obtained after centrifuging at 4000 r/min and water was added to the volume of 100 mL. Mixing 10 mL of the hydrolysate with 10 mL of 0.01 M iodine solution, 2 mL of 1 M H₂SO₄ and 18 mL of 0.1 M NaOH, then the mixture was titrated by 0.01 M Na₂S₂O₃. V₁, V₂, V₀ represents the volume titrated for A₁, A₂, B, respectively. Starch gelatinization degree was calculated by the following formula: SGD (%) = [(V₀ - V₁) / (V₀ - V₂)] \times 100

2.7. Scavenging Effects on DPPH Radical.

Briefly, 0.1 mL of the extracted solution was mixed with a methanolic DPPH solution for which the absorbance at $\lambda = 515$ nm was diluted by methanol to 0.60 \pm 0.02 [23]. The absorbance of the mixture was measured at the end of the 30 min of incubation. The scavenging effect was calculated according to the following equation:

DPPH radical scavenging rate (%) = [(A₅₁₅ control - A₅₁₅ sample) / A₅₁₅ control] \times 100, where A control = absorbance of DPPH radical + methanol, A sample = absorbance of DPPH radical + whole grain extract, and methanol was regarded as blank.

DPPH radical scavenging ability was expressed as the scavenging rate (%) per gram of whole grain (DW). Data are reported as mean \pm SD for at least three replications.

2.8. Determination of TEAC

ABTS was prepared according to Re [24]. The ABTS was then diluted with water to produce a working solution with an absorbance of 0.70 ± 0.02 at 734 nm. Each sample (0.1 mL) was mixed with 3.9 mL of working solution, and the decoloration was measured at 734 nm after 10 min. TEAC was determined by means of a proper calibration curve created by using trolox at different concentrations. Data are reported as mean \pm SD for at least three replications.

2.9. Determination of FRAP

Briefly, 0.1 mL of each sample was mixed thoroughly with 3.0 mL of FRAP reagent before the mixture was incubated for 30 min at 37°C in a water bath [25]. The absorbance was measured at $\lambda = 593$ nm. Different concentrations of FeSO_4 were used to create the calibration curve. FRAP values were obtained from the calibration curve according to the absorbance of each sample and expressed as mM FeSO_4 equivalent per gram of whole grain (DW).

2.10. Determination of Phenolic Acids by HPLC

Ferulic and *p*-coumaric acid contents in free and bound phenolic fractions were determined by HPLC [15]. Briefly, 10 μL of the extract was analyzed by HPLC (1260 Infinity, Agilent Technologies Co., Ltd., Santa Clara, USA) equipped with a G1311C quaternary pump, a G1316A column compartment, and a G1315D diode array detector. Separations were performed on a Waters LC-18 column (4.6×250 mm, 5 μm), with the column flow rate set to

0.5 mL min^{-1} and the column temperature set to 25°C. The mobile phase consisted of formic acid/water (1:99; v/v) (eluent A) and methanol/acetonitrile/formic acid (94:5:1; v/v/v) (eluent B). Gradient elution was used as follows: 0 min, 20% B; 10 min, 30% B; 15 min, 40% B; 18 min, 45% B; 20 min, 50% B; 30 min, 70% B; and 40 min, 85% B. Ferulic and *p*-coumaric acid were identified by comparing the relative retention times of eluted compounds with those of pure standards at $\lambda = 320$ nm. Ferulic and *p*-coumaric acids were quantified by an external standard method.

2.11. Statistical Analysis.

Values are shown as mean \pm SD ($n = 3$). Analysis of variance (one-way ANOVA) was used to test differences in values between treatments, with a significance level of $\alpha = 0.05$. Relationship between the variables were determined by Pearson's regression analysis. All statistical analysis were performed with SPSS software (version 19).

3. Results and Discussion

3.1. TPC of Raw and Extruded Grains

The TPC of free and bound fractions (free and bound phenolic extracts) of raw and extruded whole grains ranged from 210.5 ± 14.1 to 670.0 ± 16.4 $\mu\text{g GAE/g DW}$ and from 260.8 ± 13.9 to 1550.5 ± 42.8 $\mu\text{g GAE/g DW}$, respectively (Figure 1). There were significant differences between the raw and extruded whole grains for both free and bound fractions.

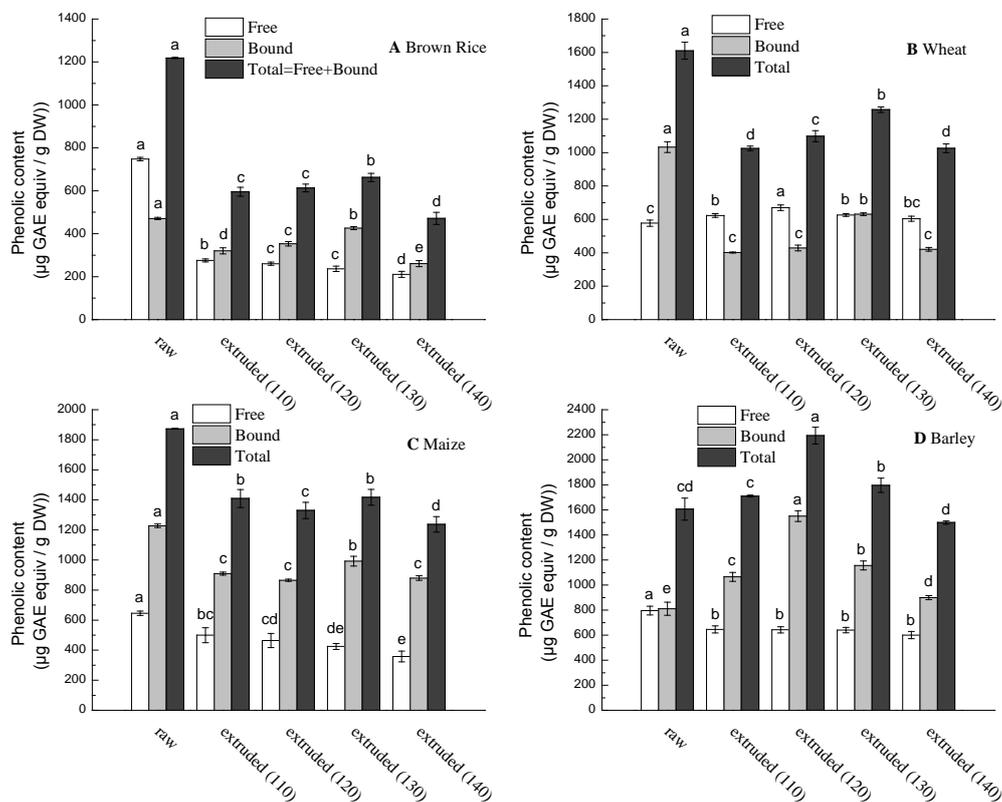


Figure 1. TPC of raw and extruded whole grains in free, bound and total fractions. Differences in values among free, bound and total fractions marked with different letters are significant ($p < 0.05$). The letter *a* represents the highest value

With respect to free fractions, the TPC of brown rice (Figure 1 A) was $747.7 \pm 8.0 \mu\text{g GAE/g DW}$ and significantly decreased ($p < 0.05$) by 63%, 65%, 68%, and 72% after extrusion (110-140°C), respectively. Similar trends were observed for maize and barley (Figure 1 B, D). However, the TPC of both maize and barley decreased less than that of brown rice under the same condition, indicating the TPC of brown rice is more susceptible than the others during extrusion. It is easy to comprehend that the TPC decreased as the temperature increased because more TPC can be decomposed at a higher temperature. Moreover, Chandrasekara [26] inferred that during thermal treatment, complexes can form between phenolics and macromolecules, making the phenolics less extractable. Interestingly, the TPC of wheat (Figure 1 B) significantly increased by 8%, 16% and 8% after extrusion (110-130°C), respectively. The increase in the free phenolic content of wheat indicates that extrusion may also have a positive effect on wheat. Dewanto [27] reported a significant increase of free TPC in sweet corn during hydrothermal treatment and attributed the increase to the liberation of the phenolic compounds from their bound forms [18].

With respect to bound fractions, the TPC of brown rice (Figure 1 A) was $470.9 \pm 5.4 \mu\text{g GAE/g DW}$ which significantly decreased by 32%, 25%, 10%, and 45% after extrusion (110-140°C), respectively ($p < 0.05$). As for wheat and maize, similar trends were also observed (Figure 1 B, C). Interestingly, TPC of barley (Figure 1 D) increased by 31%, 91%, 43%, and 11% after extrusion (110-140°C), respectively ($p < 0.05$). As is described above, phenolic compounds could be released from their bound forms after processing because of the heat. That probably is the reason why there will be a negative effect of extrusion on the bound TPC of the whole grains. It is noteworthy that a positive effect of extrusion on the bound TPC was also observed in barley. However, such a phenomenon had never been reported in previous studies [18,19,27,28]. It might be explained that extrusion may help the whole grains to be digested more thoroughly than the raw ones, which may have made the bound TPC easier to liberate and release more bound TPC during hydrolyzation. After determining the degree of starch gelatinization (Table 1), we found that the degree of starch

gelatinization increased as the temperature increased. However, no significant enhancements were found of the values after the temperature increased to the certain point. As for brown rice, wheat and maize, the temperature is 130°C. But for barley, the temperature is 120°C. Therefore, the enhancement of degree of starch gelatinization may have some effects on improving the bound TPC. Previous studies reported that the digestibility of nutrients and phytochemicals in cereal grains could be improved during processing due to the gelatinization of starch [28,29]. Compared with the other extrusion temperature, there was an apparent increase after extrusion at 130°C for brown rice, wheat and maize, and for barley at 120°C. As for brown rice, wheat and maize, the heat decreased the bound TPC at first. With the degree of starch gelatinization increased, enhancement of the bound TPC gradually compensated the loss of the bound TPC which is caused by heat. However, after the degree of starch gelatinization stopped increasing, the heat started to decrease the bound TPC again. As for barley, the degree of starch gelatinization increased the bound TPC at first because the degree of starch gelatinization reached to the maximum value at a lower temperature than the others. Similarly, the heat started to decrease the bound TPC again after the degree of starch gelatinization stopped changing.

With respect to total fractions, the change of TPC was also significant ($p < 0.05$) after extrusion (110-140°C). The total TPC of brown rice (Figure 1 A) was $1218.6 \pm 2.6 \mu\text{g GAE/g DW}$ which significantly decreased by 51%, 50%, 46%, and 61% after extrusion (110-140°C), respectively ($p < 0.05$). The same trends were also observed for the other grains (Figure 1 B, C). As for the barley, however, it increased by 7%, 37%, 12% and 7%. Correspondingly, these trends observed were similar as we observed on bound TPC indicating that the alteration occurred on total TPC is mainly decided by bound TPC, though total TPC is the combination of both free and bound TPC. This might be the reason of bound TPC's sufficiency in amount and stableness against heat. The certain temperatures on which the total TPC re-enhancement took place explained that heat and the degree of starch gelatinization could still affect the total TPC.

Table 1. Degree of starch gelatinization of extruded whole grains under different temperatures

Temperature /°C	Degree of starch gelatinization ^a / %			
	110	120	130	140
Brown rice	84.2 ± 0.8 c	91.6 ± 1.3 b	97.1 ± 0.8 a	98.0 ± 1.1 a
Wheat	85.1 ± 0.9 c	93.2 ± 0.4 b	96.7 ± 0.5 a	97.8 ± 0.6 a
Maize	82.7 ± 0.6 c	93.4 ± 1.3 b	97.4 ± 0.7 a	97.9 ± 0.6 a
Barley	91.5 ± 0.9 b	96.8 ± 1.6 a	98.1 ± 0.6 a	98.4 ± 0.8 a

^aValues in the table are all expressed as mean ± SD with 3 replications. The values in each row having the same letter are not significantly different ($p > 0.05$). The letter *a* represents the highest value in each row.

From the phenomenon above, it is reasonable to infer that the bound TPC of whole grains was spontaneously affected by heat which leads to the loss of bound TPC and degree of starch gelatinization which helps to increase the bound TPC. Perhaps the free phenols could permeate into the starch and wrapped by it during the process of starch gelatinization which leads to the protection of some free TPC from being decomposed by heat, and then be released during the extraction of bound phenols which in turn

increases the bound TPC. Bound TPC, however, could hardly increase as the temperature keep climbing up due to the higher heat could still decompose the free phenols wrapped in the starch and release the phenols from their bound form into free form. Because of bound TPC's sufficiency and stableness, the alteration of total TPC was similar to that of bound TPC which means bound TPC plays a dominant role on the alteration of the total TPC during extrusion. Therefore, extrusion can significantly

change TPC of whole grains both in free and bound fractions ($p < 0.05$). Both the heat and the degree of starch gelatinization could affect TPC of whole grains. Therefore, adopting an appropriate temperature to process whole grains can retain or improve the TPC to the highest extent.

3.2. Antioxidant Capacity of Raw and Extruded Grains

To properly evaluate the antioxidant capacity of raw and extruded whole grains, DPPH radical scavenging activity, TEAC and FRAP were used to determine the antioxidant capacity. DPPH and ABTS radicals are scavenged by antioxidants through the donation of hydrogen to form colourless compounds, and the reduction of their colour could be quantified at the wavelength of 515 nm and 734 nm. FRAP assay is used to determine the ability of the antioxidant to reduce Fe^{3+} to Fe^{2+} with the presence of TPTZ at low pH. Once Fe^{3+} is reduced to Fe^{2+} , blue compound Fe^{2+} -TPTZ could be formed and the colour could be quantified at the wavelength of 593 nm [16]. In the present study, the TEAC and FRAP assay was performed in an aqueous system, thus only the hydrophilic antioxidant capacity was measured.

In the free fraction of brown rice, the antioxidant capacity changed significantly ($p < 0.05$) after extrusion (Table 2). The results showed that over half of its initial TEAC and FRAP value were lost during extrusion (110-140°C). A similar trend was also observed in the DPPH radical scavenging ability. As for wheat, extruded wheats exhibited stronger antioxidant capacity than the raw one. Its DPPH radical scavenging ability significantly increased ($p < 0.05$) by 38%, 259%, 111%, and 98% after extrusion (110-140°C), respectively. Similar results were also obtained with the TEAC value. However, FRAP value decreased slightly after processing except for extrusion (120°C), which increased by 74% ($p < 0.05$). It is noteworthy that a remarkable increase in TEAC value, DPPH radical scavenging ability, and FRAP value was obtained after extrusion at 120°C. As for maize, both TEAC and FRAP values did not significantly change ($p > 0.05$) during extrusion (110-130°C), whereas there is a slight decrease ($p < 0.05$) after extrusion at 140°C. In contrast, the FRAP value significantly decreased after extrusion (110-140°C). As for barley, no significant increase ($p > 0.05$) was found after extrusion (110-140°C) for TEAC value. A similar trend was also shown for FRAP value. However, DPPH radical scavenging ability significantly increased.

Table 2. TEAC, DPPH, and FRAP values in the free and bound fractions of unprocessed and processed grains

Samples	TEAC ^a		DPPH ^{a,b}		FRAP ^a	
	(μmol Trolox equiv/g DW)		(Inhibition rate per gram DW)		(μmol FeSO ₄ equiv/g DW)	
	Free	Bound	Free	Bound	Free	Bound
Brown rice						
Raw	4.52 ± 0.05 a	4.97 ± 0.13 a	2.63 ± 0.07 a	2.07 ± 0.05 a	9.86 ± 0.31 a	7.03 ± 0.26 a
Extruded (110°C)	1.15 ± 0.14 c	3.53 ± 0.10 d	2.02 ± 0.09 b	1.62 ± 0.06 b	2.57 ± 0.20 e	4.84 ± 0.11 d
Extruded (120°C)	1.77 ± 0.20 b	3.91 ± 0.06 c	1.91 ± 0.07 bc	1.73 ± 0.08 b	2.95 ± 0.26 de	5.49 ± 0.12 c
Extruded (130°C)	1.83 ± 0.10 b	4.43 ± 0.06 b	1.81 ± 0.13 c	1.99 ± 0.13 a	3.46 ± 0.11 bc	6.41 ± 0.24 b
Extruded (140°C)	1.87 ± 0.25 b	3.09 ± 0.06 e	1.59 ± 0.06 d	1.42 ± 0.08 c	3.18 ± 0.12 cd	4.53 ± 0.11 d
Wheat						
Raw	3.02 ± 0.13 c	9.92 ± 0.34 a	1.08 ± 0.08 d	3.77 ± 0.06 a	5.60 ± 0.31 b	17.38 ± 0.80 a
Extruded (110°C)	3.77 ± 0.20 bc	4.37 ± 0.11 c	1.49 ± 0.11 c	2.02 ± 0.03 c	3.80 ± 0.26 c	6.56 ± 0.36 c
Extruded (120°C)	5.82 ± 0.67 a	4.43 ± 0.08 c	3.86 ± 0.09 a	2.04 ± 0.03 c	9.73 ± 0.25 a	6.73 ± 0.23 c
Extruded (130°C)	4.22 ± 0.35 b	5.80 ± 0.14 b	2.28 ± 0.05 b	2.64 ± 0.05 b	5.72 ± 0.21 b	8.88 ± 0.28 b
Extruded (140°C)	3.21 ± 0.46 c	4.55 ± 0.17 c	2.13 ± 0.13 b	2.03 ± 0.04 c	5.18 ± 0.24 b	6.75 ± 0.23 c
Maize						
Raw	3.94 ± 0.27 ab	11.55 ± 0.29 a	2.52 ± 0.08 ab	4.44 ± 0.09 a	8.40 ± 0.72 a	19.88 ± 0.38 a
Extruded (110°C)	3.89 ± 0.23 ab	9.51 ± 0.14 c	2.74 ± 0.16 a	3.88 ± 0.13 b	6.74 ± 0.36 b	14.54 ± 0.43 c
Extruded (120°C)	4.20 ± 0.28 a	9.17 ± 0.06 c	2.56 ± 0.14 ab	3.73 ± 0.05 b	6.18 ± 0.26 b	14.11 ± 0.60 c
Extruded (130°C)	4.22 ± 0.22 a	10.49 ± 0.14 b	2.42 ± 0.11 b	4.18 ± 0.18 a	5.55 ± 0.31 c	16.19 ± 0.72 b
Extruded (140°C)	3.43 ± 0.22 b	9.36 ± 0.10 c	2.16 ± 0.13 c	3.84 ± 0.13 b	4.35 ± 0.46 d	14.44 ± 0.40 c
Barley						
Raw	5.56 ± 0.17 a	7.41 ± 0.28 e	3.14 ± 0.10 b	3.01 ± 0.15 d	10.40 ± 0.44 a	10.78 ± 0.31 d
Extruded (110°C)	6.09 ± 0.63 a	10.39 ± 0.21 c	3.87 ± 0.12 a	4.83 ± 0.18 b	9.53 ± 0.70 b	15.67 ± 0.62 b
Extruded (120°C)	6.23 ± 0.32 a	13.97 ± 0.29 a	3.80 ± 0.14 a	5.86 ± 0.14 a	9.58 ± 0.53 b	21.79 ± 1.40 a
Extruded (130°C)	6.20 ± 0.21 a	11.36 ± 0.38 b	3.73 ± 0.26 a	4.87 ± 0.12 b	9.65 ± 0.24 b	16.77 ± 0.71 b
Extruded (140°C)	6.28 ± 0.36 a	8.82 ± 0.22 d	3.74 ± 0.21 a	4.13 ± 0.12 c	9.35 ± 0.43 b	13.38 ± 0.44 c

^aValues in the table are all expressed as mean ± SD with 3 replications and referred to the dry weight. ^bThe values obtained by the method described before were recalculated and referred to one gram of dried weight. The values in each column having the same letter are not significantly different ($p > 0.05$). The letter *a* represents the highest value in each column.

In the bound fraction of 4 grains, we found a similar trend of the antioxidant capacity (TEAC, DPPH radical scavenging ability and FRAP) of brown rice with its bound TPC. Compared with the results shown in Figure 1 and Table 2, the antioxidant capacity of wheat, maize, and barley was also in line with their bound TPC after

processing. Correlation analysis (Table 4) showed a statistically high positive correlation between TPC and its antioxidant capacity (for TEAC: $r = 0.988$, $p < 0.01$; for DPPH: $r = 0.978$, $p < 0.01$; for FRAP: $r = 0.979$, $p < 0.01$). The results could be explained by the fact that bound TPC is mainly responsible for the antioxidant capacity in its

bound form. The change in bound TPC could directly reflect a variation in the antioxidant capacity of whole grains.

In addition, correlation analysis between total TPC and antioxidant capacity was tested. The results (Table 4) still showed a statistically high positive correlation between total TPC and its antioxidant capacity (for TEAC: $r = 0.914$, $p < 0.01$; for DPPH: $r = 0.768$, $p < 0.01$; for FRAP: $r = 0.937$, $p < 0.01$). Though the correlation coefficient between total TPC and antioxidant capacity decreased a little for the cause of free TPC, high positive correlation between them still shows that TPC is the main antioxidant of whole grain.

To summarize, the antioxidant capacity of these four whole grains can be significantly affected by extrusion both in free and bound fractions. In the free fraction, the antioxidant capacity of extruded wheat can be well retained or improved during processing, which could be contributed by the enhancement of the free TPC in wheat because the free TPC are better extracted after extrusion along with some other antioxidants (e.g. amino acid). A previous study also found a remarkable enhancement of antioxidant capacity after processing [30]. However, brown rice lost a high proportion of its original antioxidant capacity after processing, compared with the other three grains, which may demonstrate that the antioxidants in brown rice are more vulnerable and thermally sensitive than the others. Although the FRAP,

TEAC, and DPPH assays used in this study were based on varied chemical principles, the values of DPPH radical scavenging activity were relatively higher than that of TEAC when compared with both of their original values. Because DPPH test was performed in the methanolic system instead of the aqueous system, which demonstrated that the antioxidant might exert higher antioxidant capacity in the methanolic system than aqueous system. Previous studies found lower antioxidant capacity in polar solvents because hydrogen bonding may cause dramatic changes in the H-atom donor activities of phenolic antioxidants [31M32]. As for the bound fraction, the variation in the antioxidant capacity of the 4 grains is associated with the change in its bound TPC, which indicates that the antioxidant capacity is mainly ruled by its bound TPC. Information gathered by previous studies has suggested the bound TPC may survive upper gastrointestinal digestion conditions and reach the colon to exert its health benefits locally after absorption [6,12].

3.3. Content of Phenolic Acid of Raw and Extruded Grains

As reported by previous studies [33,34], ferulic and *p*-coumaric acids are regarded as the major phenolic acids in whole grains. Table 3 shows the content of ferulic and *p*-coumaric acids in the free and the bound fractions for both processed and unprocessed whole grains.

Table 3. Content (micrograms per gram of dried weight) of ferulic acid and *p*-coumaric acid in free and insoluble bound phenolic extracts in different samples

Samples	Free ($\mu\text{g/g DW}$)			Bound ($\mu\text{g/g DW}$)		
	Ferulic acid ^a	<i>p</i> -coumaric acid ^a	Total ^{a,b}	Ferulic acid ^a	<i>p</i> -coumaric acid ^a	Total ^{a,b}
Brown rice						
Raw	49.86 \pm 0.48 a	30.60 \pm 0.73 a	80.46 \pm 1.20 a	299.03 \pm 0.98 a	145.81 \pm 3.15 a	444.84 \pm 4.12 a
Extruded (110°C)	1.66 \pm 0.01 d	2.26 \pm 0.04 d	3.93 \pm 0.05 d	177.14 \pm 2.23 d	80.48 \pm 0.12 e	257.63 \pm 2.35 d
Extruded (120°C)	9.29 \pm 0.24 b	5.50 \pm 0.06 b	14.79 \pm 0.30 b	233.29 \pm 1.94 c	101.67 \pm 0.43 c	334.96 \pm 2.36 c
Extruded (130°C)	5.00 \pm 0.07 c	1.31 \pm 0.00 e	6.31 \pm 0.08 c	264.74 \pm 1.38 b	129.67 \pm 0.31 b	394.41 \pm 1.69 b
Extruded (140°C)	0.89 \pm 0.01 e	4.02 \pm 0.17 c	4.91 \pm 0.18 d	163.33 \pm 1.10 e	85.61 \pm 0.37 d	248.93 \pm 1.48 e
Wheat						
Raw	19.47 \pm 0.24 b	12.27 \pm 0.68 a	31.73 \pm 0.91 a	735.99 \pm 0.70 a	52.79 \pm 0.75 a	788.78 \pm 0.05 a
Extruded (110°C)	4.74 \pm 0.22 d	2.07 \pm 0.02 c	6.81 \pm 0.24 d	313.66 \pm 4.28 d	40.31 \pm 0.45 c	353.96 \pm 4.74 e
Extruded (120°C)	4.39 \pm 0.16 d	3.48 \pm 0.02 b	7.87 \pm 0.18 cd	315.99 \pm 7.98 d	45.25 \pm 1.91 b	361.24 \pm 9.89 de
Extruded (130°C)	1.34 \pm 0.01 e	1.19 \pm 0.01 d	2.53 \pm 0.02 e	320.54 \pm 3.57 d	45.54 \pm 0.46 b	366.08 \pm 4.03 d
Extruded (140°C)	0.99 \pm 0.01 e	0.91 \pm 0.01 d	1.91 \pm 0.03 e	312.96 \pm 1.44 d	41.72 \pm 0.18 c	354.68 \pm 1.62 e
Maize						
Raw	37.70 \pm 0.64 a	23.82 \pm 0.90 a	61.53 \pm 1.54 a	1007.02 \pm 3.30 a	171.03 \pm 2.88 a	1178.05 \pm 6.18 a
Extruded(110°C)	12.11 \pm 0.35 b	0.78 \pm 0.01 bc	12.89 \pm 0.37 b	814.07 \pm 5.84 d	134.45 \pm 0.70 d	948.52 \pm 6.54 d
Extruded (120°C)	2.85 \pm 0.17 c	1.59 \pm 0.01 b	4.44 \pm 0.18 c	830.43 \pm 2.17 c	135.03 \pm 1.07 d	965.45 \pm 3.24 c
Extruded (130°C)	1.62 \pm 0.02 d	1.18 \pm 0.00 bc	2.80 \pm 0.03 de	906.67 \pm 3.37 b	151.76 \pm 0.35 b	1058.43 \pm 3.72 b
Extruded (140°C)	1.22 \pm 0.06 d	0.65 \pm 0.01 c	1.86 \pm 0.07 e	767.28 \pm 1.81 e	145.54 \pm 1.15 c	912.82 \pm 2.96 e
Barley						
Raw	26.81 \pm 0.41 a	16.15 \pm 0.14 a	42.96 \pm 0.27 a	418.24 \pm 4.89 d	251.56 \pm 7.58 c	669.81 \pm 2.69 d
Extruded (110°C)	3.29 \pm 0.08 d	2.81 \pm 0.01 c	6.10 \pm 0.10 d	500.46 \pm 4.49 c	252.11 \pm 0.48 c	752.56 \pm 4.97 c
Extruded (120°C)	5.01 \pm 0.12 c	2.48 \pm 0.01 d	7.49 \pm 0.13 c	795.40 \pm 4.33 a	368.94 \pm 1.11 a	1164.35 \pm 5.44 a
Extruded (130°C)	6.09 \pm 0.07 b	5.29 \pm 0.06 b	11.38 \pm 0.14 b	620.54 \pm 2.74 b	278.52 \pm 0.61 b	899.05 \pm 3.35 b
Extruded (140°C)	5.06 \pm 0.24 c	2.89 \pm 0.00 c	7.94 \pm 0.24 c	385.73 \pm 3.46 e	182.05 \pm 0.09 d	567.77 \pm 3.56 e

^aValues in the table are all expressed as mean \pm SD with 3 replications and referred to the dry weight. ^bThe total values were obtained by summing each single replicate of each compound. The values in each column having the same letter are not significantly different ($p > 0.05$). The letter *a* represents the highest value in each column.

In the free fraction of brown rice, the content of ferulic and *p*-coumaric acids was 49.9 ± 0.5 and $30.6 \pm 0.7 \mu\text{g/g DW}$. About 95%, 82%, 92%, and 94% of its total phenolic acids were lost after extrusion (110-140°C). The negative

effect of extrusion on the total phenolic acids is very notable ($p < 0.05$). A similar trend was also observed in wheat, maize, and barley. Over 70% of their total free phenolic acid were lost, which indicates some oxidation

took place during extrusion. In contrast, other authors did not find any significant decrease in total phenolic acids contents after extrusion (120-200°C) of buckwheat groat, perhaps due to different conditions and different grains [19]. A previous study has reported that phenolic acids could be released from their bound form into the free form to increase the phenolic content in the free fraction [27]. However, no increase was found for ferulic or *p*-coumaric acids after extrusion in the present study. Perhaps the free phenolic acids are very vulnerable in its free form, which could be soon decomposed after it is released from its

bound form during extrusion (110-140°C). In addition, the free phenolic acids could also permeate into the starch during extrusion leading to the decrease of their contents. Correlation analysis (Table 4) showed no significant relationship between total phenolic acid against antioxidant capacity (for TEAC: $r = 0.058$, $p > 0.05$; for DPPH: $r = 0.164$, $p > 0.05$; for FRAP: $r = 0.221$, $p > 0.05$), because most of the phenolic acids were decomposed after processing. It also indicates that the total phenolic acids are not the major components in free TPC.

Table 4. Correlation analysis of antioxidant capacity and content of antioxidant compounds of raw and extruded grain

	Total phenolic content	Total phenolic acid content ^a
Free phenolic extracts		
TEAC	0.815**	0.164 ^{n.s.}
DPPH	0.674**	0.058 ^{n.s.}
FRAP	0.823**	0.221 ^{n.s.}
Bound phenolic extracts		
TEAC	0.988**	0.951**
DPPH	0.978**	0.912**
FRAP	0.979**	0.945**
Free + Bound phenolic extracts ^b		
TEAC	0.914**	0.733**
DPPH	0.768**	0.560**
FRAP	0.937**	0.773**

** $p < 0.01$

n.s.: not significant

^aBoth ferulic acid and *p*-coumaric acid are considered.

^bBoth free and bound fractions are considered.

In the bound fraction of brown rice, the contents of ferulic and *p*-coumaric acids are 299.0 ± 1.0 and 145.8 ± 3.2 $\mu\text{g/g}$ DW, which is respectively over 6 and 4 times higher than the free fraction of the phenolic acid. Similar results from previous studies support that phenolic acid are abundant in their bound form, especially for ferulic acid and *p*-coumaric acids, which are ester-linked to cell wall polysaccharides [12,15,35]. Compared with the free phenolic acids, only about 42%, 25%, 11%, and 44% of the total phenolic acids were lost after extrusion (110-140°C), respectively. The bound fraction of wheat and maize had a similar trend during processing. Phenolic acids of barley also decreased after extrusion (-15%, $p < 0.05$) at 140°C. However, extrusion (110-130°C) significantly increased the phenolic acids content of barley. Thus, bound phenolic acids are less susceptible during processing compared with the free phenolic acids. Correlation analysis (Table 4) showed a significant relationship between total phenolic acid against antioxidant capacity (for TEAC: $r = 0.951$, $p < 0.01$; for DPPH: $r = 0.912$, $p < 0.01$; for FRAP: $r = 0.945$, $p < 0.01$). The high correlation between antioxidant capacity and total phenolic acid content also revealed the fact that ferulic acid and *p*-coumaric acid are the major components in bound TPC.

As for the whole ferulic acid and *p*-coumaric acid both in free and bound phenolic extracts, correlation analysis (Table 4) was also conducted with antioxidant capacity. Results showed a statistically high positive correlation between them (for TEAC: $r = 0.733$, $p < 0.01$; for DPPH: $r = 0.560$, $p < 0.01$; for FRAP: $r = 0.773$, $p < 0.01$). This result revealed a fact that whole phenolic acid still has a

great impact on the antioxidant though most of the phenolic acids were decomposed during extrusion which should be due to the sufficiency and stableness offered by bound phenolic acid.

4. Conclusion

In this study, we investigated changes in the TPC, phenolic acid content, degree of starch gelatinization and antioxidant capacity of different whole grains after extrusion (110-140°C).

The extrusion significantly affected the indexes above. TPC, phenolic acid content, and antioxidant capacity can be retained or even enhanced when an appropriate extrusion temperature is adopted. Extruded (120°C) wheat and barley suggests that processed whole grains may have more health benefits than the raw material.

In conclusion, the results provide useful information on how the antioxidant components, degree of starch gelatinization and antioxidant capacity changes in different whole grains during extrusion. Thus, the results might be valuable in commercial extrusion for better retention or enhancement of the antioxidants in whole grains.

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