

# Synthesis and Characterization of Biodegradable Synthetic Thickener from Anionic Triglyceride Poly(lactic acid)

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**Abstract** In this research work, we report the synthesis of novel biodegradable synthetic thickener able to provide good quality prints with reduced wastewater pollution. Biodegradable triglyceride poly(lactic acid) (TGPLA) star polymer hydrogel was prepared via one-pot reaction by direct polycondensation of lactic acid and glycerol. The chemical structure of the produced polymer was determined by Fourier transform infrared spectroscopy (FT-IR), nuclear magnetic resonance spectroscopy (<sup>1</sup>H-NMR), and gel permeation chromatography (GPC). The rheological properties and biodegradation of the novel thickening agent were investigated. The prints quality of the prepared thickening hydrogels increased with increasing lactic acid content, while the apparent viscosity and biodegradation rate increased. Technical evaluation of TGPLA as environmentally friendly thickening agent in printing of different fabrics was studied. In addition, the chemical oxygen demand (COD) and biological oxygen demand (BOD) were described. The use of such thickening composite in printing such variety of fabrics with pigment and different types of dyes including reactive, acid, metal complex, basic, and disperse dyes was described. It was established that the TGPLA composite paste is characterized by a non-Newtonian pseudoplastic performance, and their apparent viscosity increases upon neutralization. Printing pastes are very stable against storing compared to mypro-gum as natural thickener. Printing pastes thickened with TGPLA containing a ratio of 98% of lactic acid and 2% of glycerol; display similar *K/S* values and colorfastness properties compared to those prints from commercial synthetic thickeners.

**Keywords:** lactic acid, triglyceride star polymer, synthetic thickener, textile printing, biodegradability

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## 1. Introduction

The use of biodegradable additives able to provide good quality prints with reduced wastewater pollution attracted recently significant attention [1,2]. In textile printing, synthetic thickeners are used as thickening agent to avoid improper fabric handle. Polyacrylic acids synthetic thickeners are extensively used in textile printing [3,4,5]. However, polyacrylates are prepared from acrylic acid which causes severe irritation to skin and respiratory tract, while high exposure could result in pulmonary edema. Eye contact with acrylic acid also causes severe and irreversible injuries. The oral lethal dose LD<sub>50</sub> in rat is estimated as 340 mg/kg [6-11]. In addition, polyacrylic acid as synthetic thickener usually displays poor outline sharpness of prints and is not biodegradable, leading to higher wastewater pollution [4]. Furthermore, the uses of natural thickeners in textile printing cause fabric stiffness and exhibit no stability for storing [12,13,14,15]. Due to petroleum shortage, production of polymers from renewable resources has become more significant. There

has been growing attention in utilization poly(lactic acid) in a variety of applications because of its natural basis and high biodegradability. However, its implementation within textile industries has been restricted to lower temperature wet and dry processing, due to its comparatively lower glass transition temperature and melting point which results in low thermal stability and low impact resistance [16-21]. On the other hand, triglyceride oils have been utilized in the synthesis of a variety of polymers. Triglyceride oils can be considered as one of the most significant resources for biopolymers. Oil-based biopolymers possess various advantages compared to polymers from petroleum-based monomers which are not biodegradable and, in many cases, more expensive than the biodegradable oil-based biopolymers [22,23,24].

The utilization of biodegradable thickening agents presents an environmentally friendly printing practice with reduced pollution in effluent. Herein, we report the first eco-friendly TGPLA-based synthetic thickener for use in textile printing to minimize water pollution while maintaining good quality prints. Synthetic thickeners with anionic charges show great potential. We present the synthesis of novel anionic charged triglyceride poly(lactic acid)

acid synthetic thickener for textile printing on a variety of fabrics using different types of dyes. The synthesis involves polymerization reaction of lactic acid with glycerol at different ratios. The polymerization reaction gives rise to triglyceride polylactic acid of different ratios. These polymeric products will be referred to as thickening composites. Factors affecting the preparation of those composite thickeners along with their overall color fastness properties of the prints are investigated.

## 2. Experimental Section

### 2.1. Apparatus and Methods

The COD values were measured by KIT COD/1500 by POT Dichromate APP Nanocor 300D and heating block Nanocolor R-8. The TOC concentration was obtained by subtracting the inorganic carbon concentration from the total carbon concentration using Shimadzu TOC-5000 Analyzer, Japan. FTIR spectra were recorded by using an FTIR spectrophotometer (Nexus 670, Nicolet, USA) in the region of 4000-400  $\text{cm}^{-1}$  with spectra resolution of 4.0  $\text{cm}^{-1}$ . NMR spectra were recorded using a BRUKER AVANCE 400 spectrometer at 400 MHz employing deuterated DMSO- $d_6$ ; chemical shifts are reported in ppm relative to tetramethylsilane internal standard at 295°K. The color characteristics of the printed fabrics were determined on a Texflash ACS/Datacolor with a Spectraflash 600 spectrophotometer. The fastness properties were determined at Textile Research Division, National Research Centre, Egypt. The pH measurements were recorded with AD-11 ADWA digital pH meter. Molecular weight of TGPLA was measured by gel permeation chromatography (GPC, Hewlett Packard 1100 Series) using polystyrene as the standard and  $\text{CHCl}_3$  as the mobile phase.

All experimental results were collected at room temperature unless stated otherwise. Solvents used in this study were obtained from Fluka and Aldrich and were employed without further purification. Lactic acid (90%) was purchased from Aladdin Industrial Corporation. All reactions were monitored by thin layer chromatography (TLC) using Merck aluminum plates pre-coated with silica gel PF<sub>254</sub>; 20x20 cm, 0.25 mm, and detected by visualization of the plate under UV lamp (254 or 365 nm).

### 2.2. Materials and Supplies

#### 2.2.1. Commercial Thickening Agents

*Sodium alginate*. Sodium alginate of medium viscosity type manufactured by CeCa Kolloid Chemie, Paris France, was Kindly Supplied under the commercial name Ceca alginate HV/KL 600.

*Myprogum*. NP-16(Meyhall), which is a non-ionic thickening agent based on modified plant seed gum, was used at a concentration of 8%.

*Synthetic thickener*. Daico Thick 1600, synthetic thickener was kindly supplied by Daico Company, Cairo, Egypt.

#### 2.2.2. Fabrics

100% cotton, 100% polyester, 100% wool, 100% acrylic, 100% nylon, cotton/polyester (50:50), and wool/polyester

(55:45) fabrics were supplied by El-Mahalla El-Kobra Company, El-Mahalla, Egypt. The fabric were scoured in aqueous solution having a liquor ratio of 1:50 and containing 2 g  $\text{L}^{-1}$  of non-ionic detergent solution (Hostapal, Clariant, Swiss) and 2 g  $\text{L}^{-1}$  of sodium carbonate at 60°C for 30 minutes to get rid of waxes and impurities, followed by rinsing in cold water, and finally dried at room temperature.

#### 2.2.3. Dyes and Pigment

All colorants were kindly supplied by El-Mahalla El-Kobra Company, El-Mahalla, Egypt; including Realan Rot RC Granulat (Reactive Dye), Isolan Orange K-RLS 150% (Acid Dye), Suncron Blue FBL D.BL56 (Disperse Dye), Maxilon Golden Yellow GL 400% (Basic Dye), Pigment Orange 34, and Isolan Bordo S-BLN (Metal Complex Dye).

#### 2.2.4. Binder

Printofix Binder MTB was kindly supplied by Clariant.

#### 2.2.5. Other chemicals

Lactic acid, glycerol, sodium carbonate, sodium bicarbonate, ammonium persulfate, hydroquinone, and urea were of laboratory grade chemicals.

### 2.3. Synthesis of Triglyceride Polylactic Acid (TGPLA) Polymer

Triglyceride polylactic acid polymers with various molar ratios of glycerol and lactic acid were synthesized via one-step polycondensation procedure as shown in Scheme 1. The typical synthesis is proceeded as follows (for ratio: 50% LA to 50% GL): A mixture of lactic acid (18 g, 20 mmol) and glycerol (17.2 g, 20 mmol) was stirred in a 500 mL conical flask, while raising the temperature to 90°C. 2-3 drops of concentrated sulfuric acid were added. The reaction was then heated at 100°C for additional 20-30 minutes. After cooling, concentrated solution of sodium hydroxide was gradually added to neutralize the reaction mixture. The mixture was then subjected to water evaporation using rotary evaporator to obtain the pure solid content. The polymer composite can also be separated by addition of ethanol to afford a wet gel that can be filtered, dried and milled.

### 2.4. Preparation of Printing Pastes

#### 2.4.1. Acid and Metal Complex Dyes

The printing pastes were prepared according to the following recipe:

Dye	30
Urea	50
Ethylene glycol	50
Thickener	Y
Water	X
Ammonium Sulphate(1:2)	60
Ludigol	15
	-----
	1000

The amount of thickener used was either Myprogum 80 g/kg printing paste or triglyceride polyactic acid 80 g/kg.

#### 2.4.2. Disperse Dye

The printing pastes were prepared according to the following recipe:-

Dye	40
Water	X
Sodium dihydrogen Phosphate	20
Ludigol	5
Thickener	Y
	-----
	1000

The amount of thickener used was either Synthetic Thickener 30 g/kg printing paste or triglyceride polyactic acid 80 g/kg.

#### 2.4.3. Reactive Dye

Dye	30
Urea	50
Ethylene glycol	50
Thickener	Y
Water	X
Formic acid	10
Ludigol	15
	-----
	1000

The amount of thickener used was either Sodium Alginate 40 g/kg printing paste or triglyceride polyactic acid 80 g/kg.

#### 2.4.4. Basic Dye

Dye	30
Acetic Acid30%	30
Ethylene glycol	3
Thickener	Y
Hot Water	X
Citric acid	5
	-----
	1000

The amount of thickener used was either Sodium Alginate 40 g/kg printing paste or triglyceride polyactic acid 80 g/kg.

#### 2.4.5. Pigment

Pigment	40
Thickener	Y
Binder	40
Diamonum phosphate	10
Water	X
	-----
	1000

The amount of thickener used was either Synthetic Thickener 30 g/kg printing paste or triglyceride polyactic acid 80 g/kg.

## 2.5. Printing Technique

All the aforementioned printing pastes were applied to cotton, polyester, wool, acrylic, nylon, cotton/polyester, and wool/polyester fabrics using screen printing technique. Fixation of the printed garments was performed either by steaming for printed fabrics by acid, metal complex, basic, or reactive dye (at 100-103°C for 30 minutes) or thermofixation for printed fabrics by pigment and disperse dye (at 150°C for 5 minutes for pigment and 180°C for 5 minutes for disperse dye) in an automatic thermostatic oven (Wenner Mathis Co., Switzerland).

## 2.6. Washing Process

### *Washing of printed fabrics with pigment, metal complex, basic, and reactive dyes:*

Rinsing with cold water, hot water and soaping of the prints with 2g/l nonionic detergent at 60°C for 30 minutes.

### *Washing of printed fabrics with acid dye:*

Washing of printed goods was carried out as follows: Rinsing with cold water- Soaping at 30-40°C in 2g/l soap solution- Rinsing with cold water

### *Reduction clear and washing of printed fabrics with disperse dye:*

- Rinsing in cold water
  - Soaping 60 at°C with 2g/l Hostapal CV ( non ionic detergent) for 20 minutes.
  - Reduction clear with 2g/l hydrosulphite, 2g/l sodium hydroxide (32.5%), 2g/l Hostapl CV (non ionic detergent) at 40-70°C.
  - Rinsing at 60-70°C.
  - Cold rinsing.
- Finally, the samples were dried and assessed for *K/S* and fastness properties.

## 2.7. Determination of Rheological Properties

The Rheological properties of the printing pastes are expressed as the apparent viscosity at different rates of shear; were measured using Brookfield Model DV-III Programmable Rheometer.

## 2.8. Colorfastness Measurements

The color strength, expressed as *K/S* and the overall fastness properties (washing, crocking, perspiration, and light), were assessed according to the standard techniques [25,26,27,28,29].

## 2.9. Biodegradability

Biodegradability tests were carried out according to OECD Guidelines for Testing Chemicals employing procedure 301D (closed bottle test) [30]. The biological oxygen demand (BOD<sub>10</sub>) value is usually expressed in milligrams of oxygen consumed per liter of sample during 10 days of incubation at 20°C and is often used as a surrogate of the degree of organic pollution of water. BOD can be used as a gauge of the effectiveness of wastewater treatment plants.

### 3. Results and Discussion

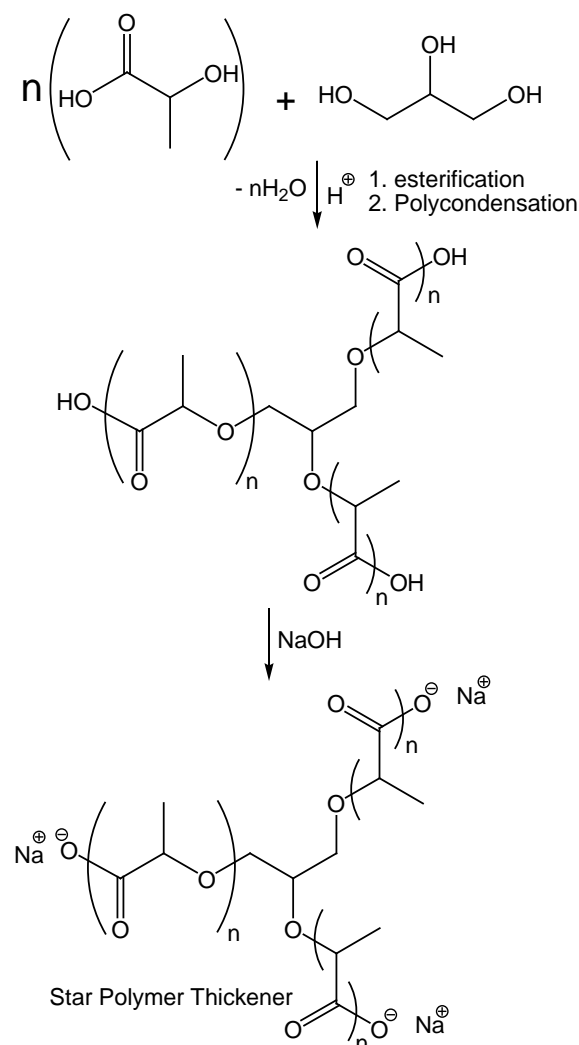
#### 3.1. Synthesis and Structural Characterization

Novel biodegradable triglyceride polylactic acid TGPLA cross-linked polymer hydrogel was prepared via one-pot reaction by thermal polycondensation of lactic acid and glycerol at different ratios in presence of sulfuric acid as a catalyst. The chemical structure of the TGPLA polymer was demonstrated by FT-IR,  $^1\text{H-NMR}$ , and GPC. The prints quality of the gels increased with increasing LA contents, while the biodegradation rate increased. LA is polymerized via polycondensation in presence of sulfuric acid catalyst, and the polymerization mechanism is shown in Scheme 1. As was expected, the properties of the obtained polymer composites depend on the concentration ratio of monomers, i.e. glycerol and lactic acid. Hence, ten different triglyceride/polylactic acid composite samples were synthesized by changing the amounts of both lactic acid and glycerol monomers as formerly stated in the experimental part. Strong broad peak at  $3426\text{ cm}^{-1}$  is assigned to the carboxylic group. Two IR peaks (Supplementary Information) at  $2988$  and  $2939\text{ cm}^{-1}$  are assigned to the absorption of the aliphatic groups including CH,  $\text{CH}_2$ , and  $\text{CH}_3$ . Peaks at  $1736$  and  $1642\text{ cm}^{-1}$  are assigned to the absorption of carbonyl ester and carbonyl amide groups, respectively, in TGPLA98. The  $^1\text{H-NMR}$  chemical shifts (Supplementary Information) at  $\delta$  4.32 and 1.12 ppm are attributed to the protons of the  $\text{C}(\text{O})\text{CH}$  and  $\text{CHCH}_3$  groups in the lactic acid units. Peaks visible at  $\delta$  1.27 and 3.59 are assigned to the central glycerol CH and the  $\text{CH}_2$  of the glycerol units respectively. These results signify the successful synthesis of the TGPLA polymer. The molar ratios of LA to GL were 98/2, 90/10, 80/20, 70/30, 60/40, 50/50, 40/60, 30/70, 20/80 and 10/90. Thus, the TGPLA samples with different compositions were abbreviated depending on the ratio of lactic acid monomer as TGPLA98, TGPLA90, TGPLA80, TGPLA70, TGPLA60, TGPLA50, TGPLA40, TGPLA30, TGPLA20 and TGPLA10, respectively. These results meant that the molecular weights of the TGPLA polymers could be controlled by adjusting the molar ratio of lactic acid to glycerol monomers. The molecular weight of TGPLA increased as longer polymerisation time proceeded. The longer polymerization time leads to high molecular weight solid state polymer. The most suitable molecular weight which was around 36,000 was obtained at 30 minutes. The TGPLA composite is miscible with water and can be separated as a soft gel by addition of ethanol as it is not soluble in water/ethanol mixture, so it can be isolated from the mixture, air-dried and finally milled. The weight average molecular weights ( $M_w$ ) of TGPLA98, from gel permeation chromatography (GPC) measurements, were controlled from 19,000 to 48,000 by changing the polycondensation reaction proceeding time as listed in Table 1.

#### 3.2. Rheological Properties of Printing Pastes

The screening started with the investigation of ten different concentration ratios of LA and GL. The

rheological properties after neutralization of the ten composite samples were investigated before and after storing for 1 and 3 days as represented in Table S2 - Table S4 respectively. It is clear that irrespective of the concentration of lactic acid monomer and/or the time elapsed before commence measuring, the relation between the shearing stress and rate of shear is not linear indicating that, these pastes are not Newtonian.



Scheme 1. Synthesis of triglyceride polylactic acid polymer

Table 1. Molecular weights of TGPLA98 copolymer upon increasing reaction time

Processing Time (min.)	$M_w$ ( $\times 10^4$ )
20	3.6
40	7.7
60	10.1
80	14.0
100	16.9

Rate of shear was ranging between 22.8 and 101.4. For example, if the rate of shear is held constant at  $21.3\text{ sec}^{-1}$ , the apparent viscosity increases from 283 to 574, 1520, to 3931 Centipoise by increasing the concentration of TGPLA from 8, 11, 14, to 17 respectively. The influence of glycerol monomer on paste rheology is investigated. It is also clear that as the concentration of lactic acid monomer increases, the apparent viscosity increases

regularly irrespective of the rate of shear. A marked reduction in the apparent viscosity curves was detected upon increasing the concentration of glycerol monomer which can be explained in terms of the increased crosslinking between glycerol and lactic acid. As the concentration of the latter raises up, the length of PLA side chains increases leading to the increment in the molecular weight, and consequently an increment in the resistance of polymer flow, i.e., an increment in apparent viscosity. On the other hand, at a constant concentration, the apparent viscosity decreases by increasing the rate of shear indicates that, these paste exhibit squaring pseudo plastic behavior. The decrease in the apparent viscosity by increasing the rate of shear means that, if the viscosity (resistance to flow) of a solution is recorded using a large applied fabric (shearing stress) which causes a high velocity of flow (shear rate), the viscosity is less than that of the same solution determined with slower rate of flow (Table 2; Figure 1 and Figure 2). In pseudo plastic solutions, no time dependence effect is observed. Furthermore, it is clear that, storing of the aforementioned pastes for 1 or 3 days at room temperature has no effect on the rheological properties of the printing paste, where it remains not Newtonian pseudo plastic. It is also obvious that the TGPLA thickener is stable for storing since the variation in the values of viscosity after storing is very marginal.

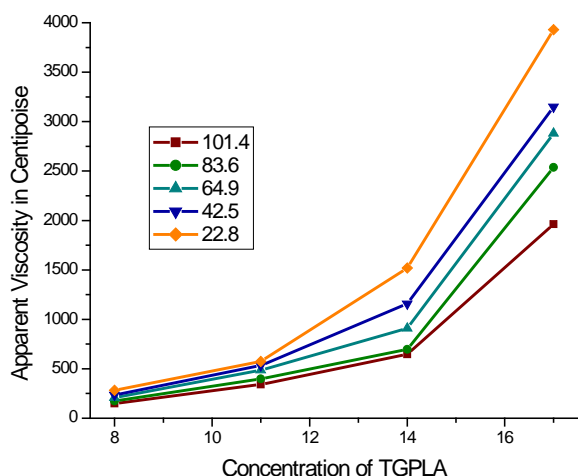
**Table 2. Effect of TGPLA polymer concentration in water on the thickening process and apparent viscosity at different rate of shear**

Concentration of TGPLA	Apparent Viscosity in Centipoise at Rate of Shear				
	22.8	42.5	64.9	83.6	101.4
8%	283	236	209	174	149
11%	574	533	486	398	341
14%	1520	1158	908	697	647
17%	3931	3150	2880	2537	1962

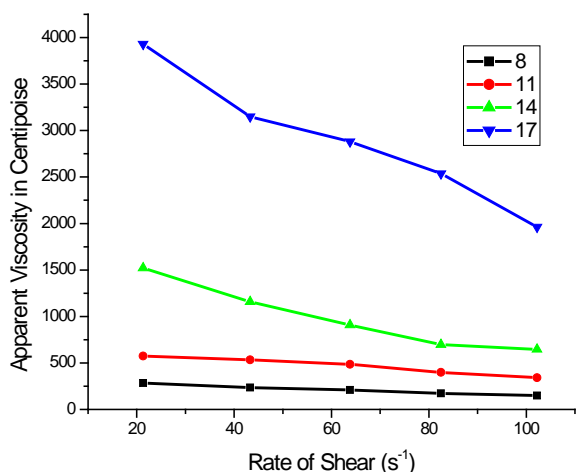
It was observed that neutralization of the TGPLA polymer composite directly after preparation has an effect on the rheological properties. The increasing in viscosity upon neutralization was surprising and may be due to the following reasons: (1) an increase in the polymer molecular weight by replacing hydrogen (atomic wt; 1) of the lactic acid carboxylic group with sodium (atomic wt; 23) to afford the sodium salt of the composite, (2) an increase in the adhesive force, or (3) an increase in the degree of formation of COO<sup>-</sup> Na<sup>+</sup> functional group more than that of the COOH carboxylic group, and consequently, the repulsion forces among the polymer chains increases. As the extent of repulsion increases, the orientation of the composite polymer molecules along the rotation axis becomes harder, and consequently, the resistance to flow increases.

### 3.3. Printing Results

To investigate the suitability of the neutralized composite (TGPLA) as thickener in printing, three different concentration ratios of LA and GL were selected (LA/GL; 10/90, 50/50, and 98/2). At similar conditions, these composites were printed on different fabrics (cotton, polyester, wool, acrylic, nylon, cotton/polyester and wool/polyester) with different colorant (Reactive, Acid, Basic, disperse, metal and pigment. For comparison, we employed additional printing pastes thickened with sodium alginate in case of reactive and Basic dye, or commercial synthetic thickener (Daico thick) in case of pigment and disperse dye or mypro gum in case of acid and metal dyes. After printing, the colored samples were air-dried and fixed by means of streaming or thermofixation as previously mentioned in the experimental section. The samples were then washed thoroughly, directed and their color strength (*K/S*) as well as their overall fastness properties was measured (Table S5 - Table S17). The use of high concentration of the glycerol monomer caused a lower paste add-on and therefore effected a small reduction in color strength but contributed to a soft fabric handle. The best fabric handle achieved using TGPLA98 and was comparable to that of prints produced with myprogum, sodium alginate or synthetic thickener. The best printing results and a very soft fabric handle were obtained when TGPLA98 at a concentration of up to 8% was used. The *K/S* of prints produced using TGPLA98 as a thickener is higher than that produced when commercial thickener was used. In addition, the prints are soft and acquire no harsh feeling. Current results suggest that the TGPLA are soluble in water and can be removed easily during washing, thereby not causing any adverse consequences on handling of the printed textile substrates. Other natural thickeners pastes are usually not appropriate for storing and are highly affected by



**Figure 1.** Effect of polymer concentration on the apparent viscosity at different rates of shear



**Figure 2.** Rheogram display dependence of the apparent viscosity on the rate of shear at different concentrations of TGPLA

fermentation; bad odor evolves upon storing for more than 24 hours and they lose most of the resistance to flow (i.e. viscosity). TGPLA can be easily washed off during the washing process. The TGPLA composites carry carboxyl groups which cannot interact with a reactive colorant and play a main role in repelling the similarly charged reactive dye ions and therefore, a high color yield is obtained.

### 3.4. Biodegradation

The biodegradation property is very important for bio-based polymers. An eco-friendly printing practice requires biodegradable printing pastes because chemicals and auxiliaries used in the printing process will be washed out following printing and fixation. A complete study on biodegradation was carried out for TGPLA composite polymers at different concentration ratios of both lactic acid and glycerol monomers. Table 3 displays the biodegradation magnitudes of these pastes. The biodegradation examinations have shown that the printing pastes made of TGPLA are more favorable than those pastes made of sodium alginate or commercial synthetic thickeners (all samples at the same concentration of  $1 \text{ g l}^{-1}$ ). This means that the use of TGPLA as thickening agent in printing pastes can lead to a significant wastewater de-pollution. Due to the high ratio of lactic acid, TGPLA98 displayed the highest biodegradable polymer with suitable apparent viscosity for printing a variety of fabrics using a variety of colorants.

**Table 3. Biodegradation analysis of commercial synthetic thickener, sodium alginate, and TGPLA polymer in aqueous medium (Conc.  $1 \text{ g l}^{-1}$ )**

Percent of LA: GL	*COD (mg O <sub>2</sub> L <sup>-1</sup> )	**BOD <sub>10</sub> (mg O <sub>2</sub> L <sup>-1</sup> )	***BOD <sub>10</sub> /COD (x 100%)
10:90%	222	62	27.9
20:80%	228	63	27.6
30:70%	231	61	26.4
40:60%	233	65	27.9
50:50%	238	68	28.6
60:40%	244	71	29.1
70:30%	253	74	29.2
80:20%	261	77	29.5
90:10%	265	80	30.2
98:02 %	274	87	31.7

\* Chemical Oxygen Demand; \*\* Biological Oxygen Demand; \*\*\* Biological Oxygen Demand in 10 days.

## 4. Conclusion

Novel aliphatic random TGPLA polymer was successfully prepared from lactic acid and glycerol monomers via polycondensation procedure using sulfuric acid as a catalyst. The produced composite is evaluated as thickening agent in textile printing. The molecular weight and composition of the TGPLA polymers could be adapted by altering the molar ratio of lactic acid and glycerol. The thickener pastes are distinguished by non-Newtonian

pseudoplastic properties, and their viscosity rises upon increasing the concentration ratio of lactic acid. The composite can be diluted by water and its apparent viscosity increases upon neutralization. Printing pastes of TGPLA composite thickener are very stable to storing. The *K/S* values and color fastness properties of prints obtained with the TGPLA thickener are equivalent to those produced by using commercial thickeners. Additionally, the use of such biodegradable thickener results in reduction of wastewater pollution. The produced polymer thickener composed of lactic acid and glycerol monomers can afford the customer with advantages in terms of environmental biocompatibility and soft handle.

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## Electronic Supplementary Information

### Synthesis, characterization and properties of biodegradable synthetic thickener from anionic triglyceride polylactic acid

Table S1. Different concentration ratios of used monomers

Percent of Lactic acid to Glycerol	Weight of Lactic Acid	Weight of Glycerol
10:90 %	3.6 g	33.12 g
20:80 %	7.2 g	29.44 g
30:70 %	10.8 g	25.76 g
40:60 %	14.4 g	22.08 g
50:50 %	18.0 g	18.4 g
60:40 %	21.6 g	14.72 g
70:30 %	25.2 g	11.04 g
80:20 %	28.8 g	7.36 g
90:10 %	32.4 g	3.68 g
98:02 %	35.28 g	0.736 g

Table S2. Effect of concentration of lactic acid on the apparent viscosity of the obtained TGPLA composite freshly prepared

Percent of Lactic/Glycerol	Apparent viscosity in Centipoise at Rate of shear ( $s^{-1}$ ) before storing (Fresh paste)				
	17.3	18.4	20.7	21.4	24.8
10:90 %	680	667	564	554	551
20:80 %	694	684	681	679	675
30:70 %	1145	1103	993	989	971
40:60 %	2580	2206	2135	2134	2131
50:50 %	2650	2413	2363	2253	2211
60:40 %	2831	2826	2369	2261	2257
70:30 %	2893	2887	2708	2673	2647
80:20 %	2984	2903	2867	2742	2663
90:10 %	3011	3001	2985	2973	2964
98:02 %	3925	3814	3766	3618	3549

**Table S3. Effect of storing for 1 day on the paste thickened with the TGPLA composite has been prepared using different concentration of Lactic Acid**

Percent of Lactic/Glycerol	Apparent viscosity in Centipoise at Rate of shear ( $s^{-1}$ ) after one day storing				
	17.3	18.4	20.7	21.4	24.8
10:90 %	679	567	549	535	498
20:80 %	688	674	667	604	598
30:70 %	1133	989	946	931	911
40:60 %	2489	1105	1103	1029	1009
50:50 %	2560	1342	1235	1233	1122
60:40 %	2793	1452	1392	1290	1236
70:30 %	2811	1703	1621	1549	1456
80:20 %	2894	1989	1975	1963	1835
90:10 %	2898	2113	2105	2019	1958
98:02 %	3997	3852	3742	3694	3589

**Table S4. Effect of storing 3 days on the paste thickened with the TGPLA composite has been prepared using different concentration of Lactic Acid**

Percent of Lactic/Glycerol	Apparent viscosity in Centipoise at Rate of shear ( $s^{-1}$ ) after three days storing				
	17.3	18.4	20.7	21.4	24.8
10:90 %	653	552	545	528	519
20:80 %	675	673	669	534	524
30:70 %	1124	1113	1110	1089	953
40:60 %	2439	1304	1204	1097	981
50:50 %	2532	2401	2330	1287	1102
60:40 %	2622	2503	2412	1331	1198
70:30 %	2731	2513	2501	1890	1345
80:20 %	2744	2613	2609	1934	1689
90:10 %	3801	3704	3676	3576	3485
98:02 %	7250	7000	6705	6511	6323

**Table S5. Comparison between K/S and fastness properties of wool fabric sample printed with prepared different TGPLA composite thickeners along with those printed using Mypro-Gum when acid used as a dye**

Thickening agent	Storing Time	K/S	Fasten properties								Light	
			Rubbing		Washing		Perspiration					
			Dry	Wet	St.	Alt	Acidic		Alkaline			
							St.	Alt.	St.	Alt.		
Mypro Gum	F	13.72	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
Mypro Gum	S	12.01	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
I	F	12.15	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
I	S	12.45	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
II	F	8.66	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
II	S	17.37	4-5	4-5	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
III	F	18.08	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
III	S	17.91	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 days, St: Staining, Alt: Alteration



**Table S6. Comparison between K/S and fastness properties of wool fabric sample printed with prepared different TGPLA composite thickeners along with those printed using Mypro-Gum when Metal used as a dye**

Thickening agent	Storing Time	K/S	Fasten properties								Light	
			Rubbing		Washing		Perspiration					
			Dry	Wet	St.	Alt	Acidic		Alkaline			
St.	Alt.	St.					Alt.					
Mypro Gum	F	6.02	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
Mypro Gum	S	6.83	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
I	F	10.35	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
I	S	7.07	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
II	F	6.35	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
II	S	6.92	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
III	F	8.94	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
III	S	8.63	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 dyes, St: Staining, Alt: Alteration.

**Table S7. Comparison between K/S and fastness properties of polyester fabric sample printed with prepared different TGPLA composite thickeners along with those printed using commercial Synthetic thickener(Dicothick) when disperse used as a dye.**

Thickening agent	Storing Time	K/S	Fasten properties								Light	
			Rubbing		Washing		Perspiration					
			Dry	Wet	St.	Alt	Acidic		Alkaline			
St.	Alt.	St.					Alt.					
Dico thick	F	17.19	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
Dico thick	S	17.26	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
I	F	17.67	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
I	S	16.47	4	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
II	F	15.39	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
II	S	16.02	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
III	F	16.80	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
III	S	16.64	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 dyes, St: Staining, Alt: Alteration.

**Table S8. Comparison between K/S and fastness properties of acrylic fabric sample printed with prepared different TGPLA composite thickeners along with those printed using commercial Synthetic thickener (Dico thick) when disperse used as a dye**

Thickening agent	Storing Time	K/S	Fasten properties								Light	
			Rubbing		Washing		Perspiration					
			Dry	Wet	St.	Alt	Acidic		Alkaline			
St.	Alt.	St.					Alt.					
Dico thick	F	23.69	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
Dico thick	S	21.55	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
I	F	22.66	4-5	4-5	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
I	S	22.28	4-5	4-5	4	4	4-5	4-5	4-5	4-5	4-5	6-7
II	F	16.16	4-5	4-5	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
II	S	17.33	4-5	4-5	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
III	F	22.33	4-5	4-5	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
III	S	22.43	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 dyes, St: Staining, Alt: Alteration.

**Table S9. Comparison between K/S and fastness properties of Nylon fabric sample printed with prepared different TGPLA composite thickeners along with those printed using commercial Synthetic thickener (Dico thick) when disperse used as a dye**

Thickening agent	Storing Time	K/S	Fasten properties								Light	
			Rubbing		Washing		Perspiration					
			Dry	Wet	St.	Alt	Acidic		Alkaline			
St.	Alt.	St.					Alt.					
Dico thick	F	22.09	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
Dico thick	S	19.60	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
I	F	17.84	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
I	S	15.58	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
II	F	16.52	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
II	S	14.67	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
III	F	21.53	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7
III	S	18.83	4-5	4	4-5	4-5	4-5	4-5	4-5	4-5	4-5	6-7

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 dyes, St: Staining, Alt: Alteration

**Table S10. Comparison between K/S and fastness properties of wool fabric sample printed with prepared different TGPLA composite thickeners along with those printed using Sodium alginate when Reactive used as a dye**

Thickening agent	Storing Time	K/S	Fasten properties								Light	
			Rubbing		Washing		Perspiration					
			Dry	Wet	St.	Alt	Acidic		Alkaline			
St.	Alt.	St.					Alt.					
Sodium alginate	F	7.59	4	4	3-4	4	3-4	4	4	4	4	6-7
Sodium alginate	S	7.37	4	4	3-4	4	4	4	4	4	4	6-7
I	F	8.29	4-5	4-5	3-4	4	4	4	3-4	4	4	6-7
I	S	8.52	4-5	4	3-4	4	4	4	4	4	4	6-7
II	F	6.41	4-5	4	3-4	4	4	4	4	4	4-5	6-7
II	S	6.23	4-5	4-5	4	4	4	4	4	4	4-5	6-7
III	F	8.63	4-5	4	3-4	4	3-4	4	4	4	4	6-7
III	S	8.52	4-5	4	4	4	4	4	4	4	4	6-7

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 dyes, St: Staining, Alt: Alteration.

**Table S11. Comparison between K/S and fastness properties of wool fabric sample printed with prepared different TGPLA composite thickeners along with those printed using Sodium alginate when Basic used as a dye**

Thickening agent	Storing Time	K/S	Fasten properties								Light	
			Rubbing		Washing		Perspiration					
			Dry	Wet	St.	Alt	Acidic		Alkaline			
St.	Alt.	St.					Alt.					
Sodium alginate	F	17.21	4-5	3-4	3-4	4	3-4	4	3-4	4	4	6-7
Sodium alginate	S	15.75	4-5	4	3-4	4	3-4	4	4	4	4	6-7
I	F	19.85	4-5	4	3-4	3-4	3	4	3-4	4	4	6
I	S	17.24	4-5	4	3-4	3-4	3	4	3-4	4	4	6
II	F	17.39	4-5	4	3-4	3-4	3	4	3	4	4	6
II	S	11.34	4-5	4	4	4	3	4	4	4	4	6
III	F	19.77	4-5	4	3-4	4	3-4	4	3	4	4	6-7
III	S	17.16	4-5	4	3-4	3-4	3	3	3-4	4	4	6-7

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 dyes, St: Staining, Alt: Alteration.

**Table S12. Comparison between K/S and fastness properties of cotton fabric sample printed with prepared different TGPLA composite thickeners along with those printed using commercial Synthetic thickener (Dico thick) when pigment used as a dye**

Thickening agent	Storing Time	K/S	Fasten properties								Light
			Rubbing		Washing		Perspiration				
			Dry	Wet	St.	Alt	Acidic		Alkaline		
St.	Alt.	St.					Alt.				
Dico thick	F	10.95	3-4	3	4	4	4-5	4	4-5	4	6
Dico thick	S	12.13	3	3	4	4	4-5	4	4-5	4	6
I	F	10.65	3-4	3	4	4	4-5	4	4-5	4	6
I	S	8.53	3-4	3	4	4	4-5	4	4-5	4	6
II	F	6.07	3-4	3-4	4	4	4-5	4	4-5	4	6
II	S	9.36	4	3	4	4	4-5	4	4-5	4	6
III	F	10.32	3-4	3	4	4	4-5	4	4-5	4	6
III	S	9.06	3	3	4	4	4-5	4	4-5	4	6

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 dyes, St: Staining, Alt: Alteration

**Table S13. Comparison between K/S and fastness properties of polyester fabric sample printed with prepared different TGPLA composite thickeners along with those printed using commercial Synthetic thickener (Dico thick) when pigment used as a dye**

Thickening agent	Storing Time	K/S	Fasten properties								Light
			Rubbing		Washing		Perspiration				
			Dry	Wet	St.	Alt	Acidic		Alkaline		
St.	Alt.	St.					Alt.				
Dico thick	F	9.19	3-4	3	4-5	4	4-5	4	4-5	4	6
Dico thick	S	8.19	3	3	4-5	4	4-5	4	4-5	4	6
I	F	9.55	3-4	3	4-5	4	4-5	4	4-5	4	6
I	S	10.17	3-4	3-4	4-5	4	4-5	4	4-5	4	6
II	F	8.75	3-4	3	4-5	4	4-5	4	4-5	4	6
II	S	8.65	3-4	3-4	4-5	4	4-5	4	4-5	4	6
III	F	9.95	3-4	3-4	4-5	4	4-5	4	4-5	4	6
III	S	8.69	3-4	3	4-5	4	4-5	4	4-5	4	6

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 dyes, St: Staining, Alt: Alteration

**Table S14. Comparison between K/S and fastness properties of cotton/polyester fabric sample printed with prepared different TGPLA composite thickeners along with those printed using commercial Synthetic thickener (Dico thick) when pigment used as a dye**

Thickening agent	Storing Time	K/S	Fasten properties								Light
			Rubbing		Washing		Perspiration				
			Dry	Wet	St.	Alt	Acidic		Alkaline		
St.	Alt.	St.					Alt.				
Dico thick	F	11.96	3-4	3	4-5	4	4-5	4	4-5	4	6
Dico thick	S	12.93	3-4	3-4	4-5	4	4-5	4-5	4-5	4	6
I	F	9.06	3-4	3	4-5	4	4-5	4	4-5	4	6
I	S	12	3-4	3	4-5	4	4-5	4	4-5	4	6
II	F	10.12	4	4	4-5	4	4-5	4	4-5	4	6
II	S	8.13	3-4	3-4	4-5	4	4-5	4	4-5	4	6
III	F	10.78	3-4	3	4-5	4	4-5	4	4-5	4	6
III	S	8.8	3-4	3-4	4-5	4	4-5	4	4-5	4	6

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 dyes, St: Staining, Alt: Alteration

**Table S15. Comparison between K/S and fastness properties of acrylic fabric sample printed with prepared different TGPLA composite thickeners along with those printed using commercial Synthetic thickener (Dico thick) when pigment used as a dye.**

Thickening agent	Storing Time	K/S	Fasten properties								Light
			Rubbing		Washing		Perspiration				
			Dry	Wet	St.	Alt	Acidic		Alkaline		
St.	Alt.	St.					Alt.				
Dico thick	F	12.07	3-4	3	4-5	4	4-5	4	4-5	4	6
Dico thick	S	11.86	3	3	4-5	4	4-5	4	4-5	4	6
I	F	11.68	3-4	3	4-5	4	4-5	4	4-5	4	6
I	S	11.95	3	3	4-5	4	4-5	4	4-5	4	6
II	F	7.48	3-4	3	4-5	4	4-5	4	4-5	4	6
II	S	6.88	3	3	4-5	4	4-5	4	4-5	4	6
III	F	12.69	3	3	4-5	4	4-5	4	4-5	4	6
III	S	11.11	3	3	4-5	4	4-5	4	4-5	4	6

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 dyes, St: Staining, Alt: Alteration

**Table S16. Comparison between K/S and fastness properties of polyester/wool fabric sample printed with prepared different TGPLA composite thickeners along with those printed using commercial Synthetic thickener (Dico thick) when pigment used as a dye**

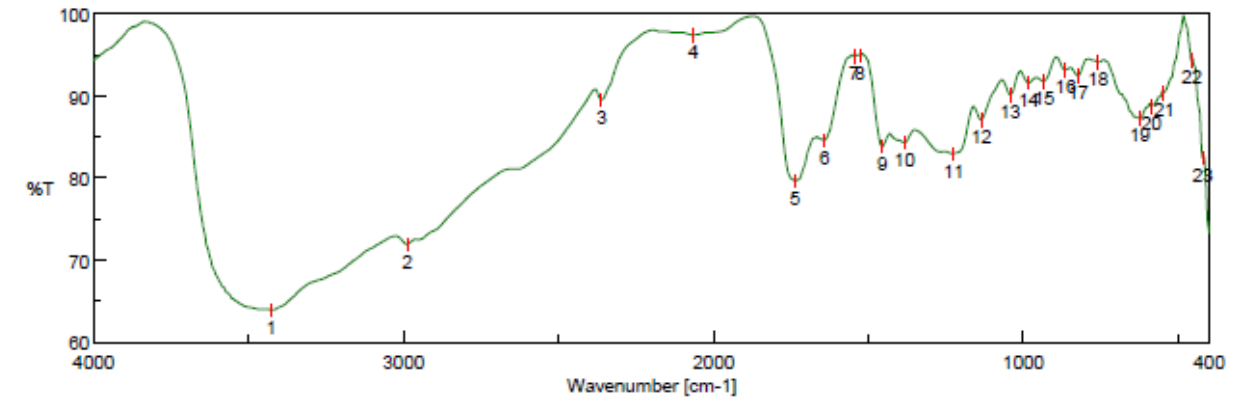
Thickening agent	Storing Time	K/S	Fasten properties								Light
			Rubbing		Washing		Perspiration				
			Dry	Wet	St.	Alt	Acidic		Alkaline		
St.	Alt.	St.					Alt.				
Dico thick	F	10.79	3-4	3	4-5	4	4-5	4	4-5	4	6
Dico thick	S	10.42	3	2-3	4-5	4	4-5	4	4-5	4	6
I	F	11.62	3-4	3	4-5	4	4-5	4	4-5	4	6
I	S	7.34	3	3	4-5	4	4-5	4	4-5	4	6
II	F	6.91	4	3-4	4-5	4	4-5	4	4-5	4	6
II	S	6.66	4	3	4-5	4	4-5	4	4-5	4	6
III	F	9.75	3-4	3	4-5	4	4-5	4	4-5	4	6
III	S	8.25	3	2-3	4-5	4	4-5	4	4-5	4	6

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 dyes, St: Staining, Alt: Alteration

**Table S17. Comparison between K/S and fastness properties of nylon fabric sample printed with prepared different TGPLA composite thickeners along with those printed using commercial Synthetic thickener (Dico thick) when pigment used as a dye.**

Thickening agent	Storing Time	K/S	Fasten properties								Light
			Rubbing		Washing		Perspiration				
			Dry	Wet	St.	Alt	Acidic		Alkaline		
St.	Alt.	St.					Alt.				
Dico thick	F	6.77	4	2-3	4-5	4	4-5	4	4-5	4	6
Dico thick	S	5.64	3-4	3-4	4-5	4	4-5	4	4-5	4	6
I	F	9.45	4	3	4-5	4	4-5	4	4-5	4	6
I	S	9.16	3	3	4-5	4	4-5	4	4-5	4	6
II	F	5.32	4	3	4-5	4	4-5	4	4-5	4	6
II	S	6.60	3	3	4-5	4	4-5	4	4-5	4	6
III	F	7.94	4	3	4-5	4	4-5	4	4-5	4	6
III	S	5.98	4	3	4-5	4	4-5	4	4-5	4	6

I, II, and III are TGPLA synthetic thickeners prepared at 10, 50, and 98 % of lactic acid, respectively. F: Freshly prepared, S: Stored for 3 dyes, St: Staining, Alt: Alteration



[ Result of Peak Picking ]

No.	Position	Intensity	No.	Position	Intensity	No.	Position	Intensity
1	3426.89	63.9239	2	2988.16	71.9106	3	2361.41	89.52
4	2064.42	97.4154	5	1736.58	79.6095	6	1642.09	84.5548
7	1546.63	94.8193	8	1528.31	94.9206	9	1454.06	83.8397
10	1383.68	84.284	11	1226.5	82.9444	12	1135.87	87.0839
13	1042.34	90.1428	14	982.554	91.5899	15	934.342	91.7895
16	863.953	93.1496	17	823.455	92.4366	18	759.816	94.1084
19	624.823	87.311	20	587.218	88.7355	21	546.72	90.3779
22	457.047	94.2657	23	421.37	82.4471			

Figure S1. FT-IR spectra of TGPLA98

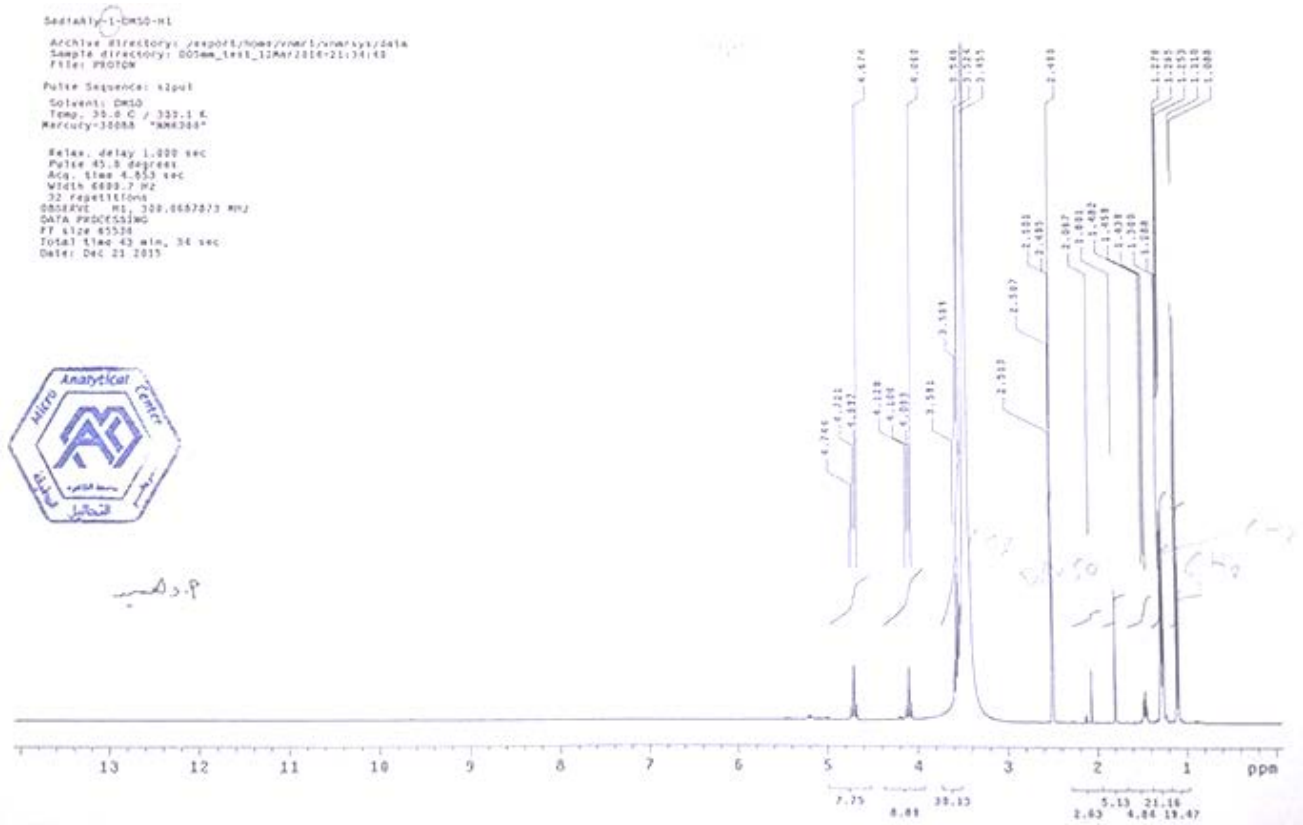


Figure S2. <sup>1</sup>H NMR spectra of TGPLA98